

# SEARCH REPORT



STIC Database Tracking Number: 316187

To: BROOKE PURINTON

Location: JEF-3B15 Art Unit: 2881

Friday, December 04, 2009

Case Serial Number: 10/599555

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# Search Notes

Attached are the edited search results and the search histories from searching EAST databases and CAS/STN Chemical Abstracts.

The search histories are included at the end of this file.

I recommend that you review at least Page 2 to about Page 20 of this file.

After the first 20 pages: The second half of this file has some abstracts about the existence and properties of the isomers explicitly mentioned in the claims.

An apparently comprehensive list of isomers and their half lives is at the "Nuclear Physics A" journal article at this URL: <a href="http://amdc.in2p3.fr/nubase/Nubase2003.pdf">http://amdc.in2p3.fr/nubase/Nubase2003.pdf</a>
I've included some data excerpts in the search results.

If you would like more searching on this case, or if you have questions or comments, please notify me.

DERWENT-ACC-NO: 1994-250295 COPYRIGHT DERWENT INFORMATION LTD

TITLE: Vasculer implant containing two or more radionuclides with specific half lives for long lasting prevention of restinosis or short lived isotope disintegrating to long lived daughter

INVENTOR: FEHSENFELD P; HEHRLEIN C

PRIORITY-DATA: 1993DE-4315002 (May 6, 1993)

#### PATENT-FAMILY:

PUB-NO	PUB-DATE	LANGUAGE
DE 4315002 C1	August 18, 1994	DE
WO 9426205 A1	November 24, 1994	DE
EP 696906 A1	February 21, 1996	DE
JP 08508436 W	September 10, 1996	JA
EP 696906 B1	April 16, 1997	DE
<u>US 5674177</u> A	October 7, 1997	EN
JP 2735689 B2	April 2, 1998	JA

ABSTRACTED-PUB-NO: DE 4315002 C1

Vascular implant for inhibiting and/or eliminating vascular constrictions contains at least two radionuclides, one with half life 7 hrs. to 7 days, the other with half life over 100 days.

Alternatively, the implant contains a single isotope of half life 7 hrs. to 7 days that disintegrates directly or indirectly to a daughter nuclide of half life over 100 days.

Pref. nuclides are esp. gamma emitters, esp. for short half lives Co 55, Mn 52, Tc 96, Mo 99 and Ni 57 and for long half lives Fe 55, Zn 65 or Co 57.

USE/ADVANTAGE - These implants (stents) are used to prevent restenosis. The entire region where there is a risk of restenosis is irradiated for the entire period that the stent is in place. Initially, the radiation dose is fairly high, subsequently it is moderate but relatively long lasing. The radionuclides can be generated in the implant after this has been made, i.e. they are not applied as a coating which might become detached.

#### US 5674177 Abstract Text - ABTX (1):

In a vascular implant for the prevention or elimination of vascular restrictions a tubular body to be inserted into a body vessel includes at least a first nuclide species which has a half life in the range of 7 hours to 7 days and a second nuclide species which has a half life of more than 100 days. Instead of providing a second nuclide species, the first nuclide species may also decay into the second nuclide species thereby providing a high initial radioactivity for a relatively short period and a relatively low radioactivity over a relatively long period.

DERWENT-ACC-NO: 1997-153036 COPYRIGHT DERWENT INFORMATION LTD

TITLE: Marking and identification of articles comprises use of radioactive isotopes with
different half-life periods, applied in determined ratio
to article

INVENTOR: LYAPIDEVSKII V K

PRIORITY-DATA: 1993RU-026033 (May 20, 1993)

PATENT-FAMILY:

PUB-NO PUB-DATE LANGUAGE

RU 2064697 C1 July 27, 1996 RU

An identifying marker is applied to an article being marked and contains radioactive substances. The radioactivities of the substances are registered and are used for the identification of the article. A mixture of radioactive isotopes with different half-life periods is used and their ratios in the identifying marker are varied, so that the number of decays of each isotope is from 100-1000000 decays per sec.. The summed amplitude distribution is measured and acts as a code of the article at the moment of its registration. The code and the date of registration are fixed in a long-term memory and are used to identify the article at any moment of time taking into account the change of the amplitude distribution because of partial decay of the radioactive isotopes in the identifying marker.

USE - Used for the marking and identifying of articles.

ADVANTAGE - The process gives a reduced complexity of application of radioactive substances.

#### (57) Abstract:

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FIELD: radiation monitoring. SUBSTANCE: object is marked with an identification sign containing radioactive substances, and its radioactivity is measured and is used for recognizing of the object. Mixtures of

isotopes with different half-life periods are used. The object is identified according to the pulse-height distribution with due account made for partial decay of radioactive isotopes. EFFECT: facilitated procedure.

#### From RU 2064697 C1, Page 3:

2. Способ по п. 1, отличающийся тем, что опознавательный знак наносят на объект в виде слоя элоксидной смолы, содержащей радиоактивные изотопы  $Na^{22}$ ,  $Co^{60}$ ,  $Nb^{96}$ ,  $Sb^{125}$ ,  $Cs^{134}$ ,  $Cs^{137}$ ,  $Bi^{207}$ , их смеси и соединения.

US-PAT-NO: 5782742 DOCUMENT-IDENTIFIER: US 5782742 A

\*\*See image for Certificate of Correction\*\*

TITLE: Radiation delivery balloon

DATE-ISSUED: July 21, 1998

----- KWIC -----

Detailed Description Text - DETX (34):

The activity of the material that is produced is a function of the new isotope's half-life and irradiation time. The activity A, produced is determined by the formula . . . . Detailed Description Text - DETX (35):

in which e is the base of the Naperian or natural logarithm (i.e., 2.718 . . .), and the elapsed time is t. .lambda. designates the decay constant of the isotope that is formed, and is calculated as the Naperian logarithm of two divided by the half life of the isotope. (It is important that time is measured in the same units in all of these calculations.) It should be noted that after a given time, a saturation point is reached, and no further activity is produced by neutron bombardment, i.e., the material is decaying away as fast as it is being produced. This represents an upper limit for the specific activity of a given isotope that can be produced under these conditions.

Detailed Description Text - DETX (52):

As an alternative to P-32, other radionuclides such as Yttrium-90 (Half-life=64.0 hours; Maximum Beta Energy=2.27 Mev; Average Beta Energy=0.9314 MeV; Betas Per Nuclear Disintegration=1.00; Maximum Range in Tissue=1.11 cm); Gold-198 (Half-life=2.696 Days; Maximum Beta Energy=961 keV; Average Beta Energy=316.3 keV; Betas Per Nuclear Disintegration=1.00; Maximum Range in Tissue 0.42 cm); and Iridium-192 (Half-life=74.2 Days; Maximum Beta Energy = 666 keV; Average Beta Energy=222 keV; Maximum Range in Tissue=0.25

cm) may also be useful.

#### US 5782742 Claim:

23. A radiation delivery balloon catheter as in claim 21, wherein the radiation delivery source is selected from the group consisting of Pt. P-32. Y-90. Au-198. Ir-192. Mo-99, and combinations thereof.

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FYI: data at <a href="http://amdc.in2p3.fr/nubase/Nubase2003.pdf">http://amdc.in2p3.fr/nubase/Nubase2003.pdf</a></a>
<a href="Gata">G. Audi et al. / Nuclear Physics A 729 (2003) 3-128</a>

Nuclide	Mass: excess (keV)		****	teitation rgy(keV)	Half-life	J <sup>™</sup> Ens		Reference	Decay modes and intensities (%)		
192 <sub>13</sub> , 192 <sub>13</sub> , 192 <sub>13</sub> ,	-34833.2 -34776.5 -34665.1	1.7 1.7 1.7	56.720 168.14	0.005 0.12	73.827 d 0.013 1.45 m 0.05 241 v 9	4 <sup>+</sup> 1 <sup>-</sup> (11 <sup></sup> )	98 98 98		β'''=95.13 14; ε=4.87 14 IT≈100; β'''=0.0175 IT=100		
<sup>98</sup> Au	-29582.1 -29269.9	0.6 0.6	312,2200	0.0020	2.69517 d 0.0002 124 ns 4		02 02		β=160 IT=100		
	-85965.8 -85868.0	1.9 1.9	97.785	0.003	65.94 h 0.01 15.5 μs 0.2	$1/2^{+}$ 9 $5/2^{+}$ 9	_		β=100 IT=100		

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US-PAT-NO: 5855546

DOCUMENT-IDENTIFIER: US 5855546 A

TITLE: Perfusion balloon and radioactive wire delivery system

DATE-ISSUED: January 5, 1999

INVENTOR-INFORMATION:

NAME CITY STATE ZIP CODE COUNTRY Hastings; Roger N. Maple Grove MN N/A N/A Urick; Michael J. Rogers MN N/A N/A

----- KWIC -----

Detailed Description Text - DETX (52):

As previously stated, a preferred source of radiation for all embodiments of the present invention is the radioactive compound Nickel-66. Nickel-66 decays with a half life of 2.28 days with only low energy beta emissions and no gamma emission into its daughter element Copper-66. Copper-66 then emits high energy beta radiation with a half life of 5.10 minutes and decays into the stabile element Zinc-66. This two-step decay has a particular advantage in use in the catheters of the present invention.

Detailed Description Text - DETX (53):

The Nickel-66 acts as a carrier for the high energy copper decay allowing for time to transport the source to the end user, and also allows for disposal of the device through ordinary means in about 23 days. A Copper-66 source alone would decay quickly and not be useful without the parent Nickel. Nickel is low cost and has desirable mechanical properties in its pure form and in alloys, such as a Nickel Titanium alloy.

Detailed Description Text - DETX (55):

Another preferred radiation source is Gadolinium-153. Gadolinium-153 is a composite gamma source which can provide low energy gammas to vessel intima layer while providing higher energy gammas to penetrate calcified plaques and reach the adventitia. . . .

FYI: data at http://amdc.in2p3.fr/nubase/Nubase2003.pdf

G. Audi et al. / Nuclear Physics A 729 (2003) 3–128

Nuclide	Mass excess (keV)		Excitation energy(keV)	]	Half-life			Ens Reference		Decay modes and intensities (%)		
66Ni	-66006.3	1.4		54.6	ħ	0.4	9+	98		β =100		
<sup>66</sup> Cu	-66258.3	0.7		5.120	m	0.914	1+	98		β ==100		
84			G. Audi et a	ıl. / Nucle	ear F	hysics A	7397	2003	03-128			

Nuclide	Mass exc (keV)	ess	454-	ritation gy(keV)	)	Half-l	ife	<i>J</i> **	Ens	Reference	Decay modes and intensities (%)
123 G4m	72889.8 72794.6 72718.6	2.5 2.5 2.5	95.1737 171.189	0.0012 0.005	240.4 3.5 76.0	d µs µs	1.0 0.4 1.4	3/2" (9/2") (11/2")			ε=100 IT=100 IT=100

US-PAT-NO: 5873811 DOCUMENT-IDENTIFIER: US 5873811 A

TITLE: Composition containing a radioactive component for treatment of vessel wall

DATE-ISSUED: February 23, 1999

INVENTOR-INFORMATION:

NAME CITY STATE ZIP CODE COUNTRY Wang; Lixiao Maple Grove MN N/A N/A Hastings; Roger N. Maple Grove MN N/A N/A

#### CLAIMS:

What is claimed:

 A method for inhibiting restenosis at a vessel wall region comprising the steps: providing an adhesive including a radioactive material; and applying said adhesive to said vessel wall region.

- 2. A method as recited in claim 1 wherein said radioactive material is admixed with said adhesive.
- 3. A method as recited in claim 1 wherein said radioactive material is chemically bonded to said adhesive.
- 4. A method as recited in claim 1 wherein said radioactive material is selected from the group consisting of: Phosphorus 32, Yttrium 90, Iodine 125, Iridium 192, and mixtures thereof.

[Note, US 6422989 has wording similar to that in US 6019718.]

US-PAT-NO: 6019718 DOCUMENT-IDENTIFIER: US 6019718 A

TITLE: Apparatus for intravascular radioactive treatment

DATE-ISSUED: February 1, 2000

INVENTOR-INFORMATION:

NAME CITY STATE ZIP CODE COUNTRY Hektner; Thomas R. Medina MN N/A N/A

----- KWIC -----

Detailed Description Text - DETX (28):

As previously stated, a preferred source of radiation for all embodiments of the present invention is the radioactive compound Nickel-66. Nickel-66 decays with a half life of 2.28 days with only low energy beta emissions and no gamma emission into its daughter element Copper-66. Copper-66 then emits high energy beta radiation with a half life of 5.10 minutes and decays into the stabile element Zinc-66. This two-step decay has a particular advantage in use in the catheters of the present invention.

Detailed Description Text - DETX (29):

The Nickel-66 acts as a carrier for the high energy copper decay allowing for time to transport the source to the end user, and also allows for disposal of the device through ordinary means in about 23 days. A Copper-66 source alone would decay quickly and not be useful without the parent Nickel. Nickel is low cost and has desirable mechanical properties in its pure form and in alloys, such as a Nickel Titanium alloy.

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US-PAT-NO: 6045495 DOCUMENT-IDENTIFIER: US 6045495 A

TITLE: Apparatus and method to treat a disease process in a luminal structure

DATE-ISSUED: April 4, 2000

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The radiation source may be a pellet, a wire, an encapsulated radiation source, or an attached radiation source, such as a paste of Ir-192, I-125, or Pd-103. Alternatively, the 10 radiation source may be a γ-radiation emitting isotope, such as, for example, one of the following: <sup>109</sup>Cd, <sup>113</sup>Sn, <sup>125</sup>Te, <sup>125</sup>I, <sup>93</sup>MO, <sup>133</sup> Ba, <sup>145</sup>Sm, <sup>147</sup>Eu, <sup>146</sup>Gd, <sup>157</sup>Tb, <sup>254</sup>Es, <sup>242</sup>Am, <sup>169</sup>Yb, <sup>186</sup>Re, <sup>173</sup>Lu, <sup>172</sup>Hf, <sup>177</sup>Lu, <sup>179</sup>Hf, <sup>183</sup>Re, <sup>44</sup>Tl, <sup>178</sup>Hf, <sup>57</sup>Co, <sup>178</sup>Hf, <sup>57</sup>Co, <sup>161</sup>Rh, <sup>75</sup>Se, <sup>123</sup>Te, <sup>139</sup>Ce, <sup>15</sup> 166Ho, 235U, <sup>101</sup>Rh, <sup>168</sup>Tm, <sup>176</sup>Lu, <sup>127</sup>Xe, <sup>95</sup>Te, <sup>177</sup>Lu, <sup>121</sup>Te, <sup>210</sup>Bi, <sup>182</sup>Hf, <sup>203</sup>Hg, <sup>175</sup>Lu, <sup>192</sup>Ir, <sup>194</sup>Ir <sup>150</sup>Eu, <sup>175</sup>Hf, <sup>249</sup>Cf, <sup>88</sup>Zr, <sup>75</sup>Se, <sup>210</sup>Bi, <sup>182</sup>Hf, <sup>203</sup>Hg, <sup>176</sup>Lu, <sup>178</sup>Hf, <sup>95</sup>Te, <sup>121</sup>Te, <sup>203</sup>Hg, <sup>192</sup>Ir <sup>178</sup>Hf, <sup>150</sup>Eu, <sup>249</sup>Cf, <sup>88</sup>Zr.

FYI: data at http://amdc.in2p3.fr/nubase/Nubase2003.pdf

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Nuclide Mass excess Excitation Half-life J<sup>R</sup> Ens Reference Decay modes and (keV) energy(keV) intensities (%)

1/8Hf.	-52444.3	2.1			STABLE			0.	94			18=27.28.7
$^{178}\mathrm{Hfm}$	-51296.9	2.1	1147.423	0.005	4.0	5	0.2	8	94			IT=100
178 Hin	-49998.6	2.1	2445.69	0.11	31	У	ž	16 <sup>+</sup>	94	94Ki.A	E	IT=100
178 Hip	-49870.8	2.2	2573.5	0.5	68	$\mu s$	2	$(14^{-})$	94			IT=100

US-PAT-NO: 6187037

DOCUMENT-IDENTIFIER: US 6187037 B1

TITLE: Metal stent containing radioactivatable isotope and

method of making same

DATE-ISSUED: February 13, 2001

INVENTOR-INFORMATION:

NAME CITY STATE ZIP CODE COUNTRY Satz; Stanley Surfside FL 33154 N/A

#### ABSTRACT:

A metal stent for vascular implantation comprising a generally tubular structure whose external surface is adapted to engage the interior vascular surface when implanted, said metal of said stent containing a substantially uniform dispersion of from about 0.05 to about 10.00 percent by weight of one or more naturally occurring or enriched stable isotopes having a half-life of less than two months and that are principally beta particle emitters, so that when activated, said stent emits low to moderate dosages of radiation uniformly to reduce cell proliferation.

#### US 6187037 Claims Text - CLTX (10):

10. The stent as described in claim 1 wherein said radio-activatable isotope results in a radioactive isotope selected from the group consisting of antimony-120, antimony-127, barium-128, barium-131, barium-140, bromine-80m, cadmium-115, cerium-134, cerium-141, cerium-143, cobalt-55, copper-64, copper-67, dysprosium-166, erbium-172, gallium-166, gallium-68, germanium-71, gold-198, gold-199, iodine-124, iodine-125, iodine-131, iridium-194, lanthanum-140, lutetium-172, neodymium-140, nickel-66, niobium-95, osmium-191, palladium-100, phosphorus-32, phosphorus-33, platinum-188, platinum-191, platinum-193m, platinum-195m, platinum-197, praseodymium-143, rhenium-186, rhenium-188, rhodium-99, rhodium-101m, rhodium 103m, rhodium-105, rubidium-82, ruthenium-103, scandium-48, silver-111, strontium-82, tantalum-177, tantalum-183, tellurium-132, tellurium-118, terbium-153, terbium-156, thallium-201, thallium-204, thulium-170, thulium-172, tin-117m, tin-121, titanium-45, tungsten-178, ytterbium-166, ytterbium-169, yttrium-87, yttrium-90, yttrium-91, zinc-72, zirconium-89, and mixtures thereof.

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[Note, US 6652441 has wording similar to that in US 6217503.]

US-PAT-NO: 6217503

DOCUMENT-IDENTIFIER: US 6217503 B1

TITLE: Apparatus and method to treat a disease process in a luminal structure

DATE-ISSUED: April 17, 2001

#### From US 6217503 Column 10:

The radiation source may be a pellet, a wire, an encapsulated radiation source, or an attached radiation source, such as a paste of kr-192, I-125, or Pd-103. Alternatively, the radiation source may be a γ-radiation emitting isotope, such as, for example, one of the following: <sup>109</sup>Cd, <sup>123</sup>Sn, <sup>125</sup>Te, <sup>125</sup>I, <sup>93</sup>Mo, <sup>123</sup>Ba, <sup>145</sup>Sm, <sup>147</sup>Eu, <sup>146</sup>Gd, <sup>157</sup>Tb, <sup>254</sup>Es, <sup>242</sup>Am, <sup>169</sup>Yb, <sup>185</sup>Re, <sup>173</sup>Lu, <sup>172</sup>Hf, <sup>177</sup>Lu, <sup>179</sup>Hf, <sup>185</sup>Re, <sup>44</sup>Tl, <sup>178</sup>Hf, <sup>57</sup>Co, <sup>178</sup>Hf, <sup>57</sup>Co, <sup>101</sup>Rh, <sup>75</sup>Se, <sup>123</sup>Te, <sup>129</sup>Ce, <sup>166</sup>Ho, <sup>235</sup>U, <sup>101</sup>Rh, <sup>166</sup>Tm, <sup>176</sup>Lu, <sup>127</sup>Xe, <sup>95</sup>Te, <sup>177</sup>Lu, <sup>121</sup>Te, <sup>25</sup>Zi<sup>5</sup>Bi, <sup>182</sup>Hf, <sup>203</sup>Hg, <sup>176</sup>Lu, <sup>102</sup>Ir, <sup>134</sup>T, <sup>150</sup>Eu, <sup>175</sup>Hf, <sup>249</sup>Cf, <sup>88</sup>Zi, <sup>75</sup>Se, <sup>210</sup>Bi, <sup>182</sup>Hf, <sup>202</sup>Hg, <sup>176</sup>Lu, <sup>175</sup>Hf, <sup>95</sup>Te, <sup>121</sup>Te, <sup>203</sup>Hg, <sup>192</sup>Ir, <sup>176</sup>Hf, <sup>150</sup>Eu, <sup>249</sup>Cf, <sup>88</sup>Zz.

#### From US 6217503 Column 17:

The radioisotopes that decay with emission of beta plus or beta mimus radiation, that have a half-life of between approximately 1 and 72 hours, that have an average decay energy of approximately 500–2000 keV, and that have radiation intensity of greater than or equal to approximately 50%, said radiation intensity being measured in % per decay, may be selected from the group consisting of NA-24, SI-31, K-42, SC-43, SC-44, CO-55, MN-56, CU-61, NI-65, 20 GA-66, GA-68, ZN-71, GA-72, AS-72, SE-73, BR-75, AS-76, BR-76, GE-77, KR-77, AS-78, Y-85, KR-87, ZR-87, NB-89, Y-90, NB-90, SR-91, Y-92, Y-93, ZR-97, IN-111, AG-112, AG-113, SB-122, SN-127, TE-129, BA-139, LA-140, LA-141, LA-142, PR-142, PR-145, TB-148, 25 PM-150, EU-152, HO-166, RE-188, RE-190, IR-194, BI-212, and radioactive sodium-chioride.

From US 6217503 Column 34, Line 48:
Further, isotopes that decay to those listed in the application may be used.

See also US 6217503 Tables 1-4.

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#### FYI: data at http://amdc.in2p3.fr/nubase/Nubase2003.pdf

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Nuclide	Mass excess (keV)	Excitation energy(keV)	Half-life	$J^{\pi}$	Ens Reference	Decay modes and intensities (%)

<sup>178</sup> Hf	-52444.3	2.1			STABLE			0+	94			IS=27.28 7
178 H2m	-51296.9	2.1	1147,423	0.005	4.0	5	0.2	8	94			IT=100
178 H±1	-49998.6	2.1	2445.69	0.11	31	У	3	16+	94	94Ki.A	E	IT=100
<sup>178</sup> H£¤	-49870.8	2.2	2573.5	0.5	68	11.5	2	$(14^{-})$	94			IT=100

[Note, US 6287249, US 6458069, and US 6491619 have wording similar to that in US 6261320.]

US-PAT-NO: 6261320 DOCUMENT-IDENTIFIER: US 6261320 B1

\*\*See image for Certificate of Correction\*\*

TITLE: Radioactive vascular liner

DATE-ISSUED: July 17, 2001

INVENTOR-INFORMATION:

NAME CITY STATE ZIP CODE COUNTRY Tam; Lisa A. Lake Forest CA N/A N/A Trauthen; Brett A. Newport Beach CA N/A N/A

From Column 27, Line 12:

One preferred embodiment of radioactive-coated stent of the present invention is that which has an isotope layer comprising the gamma-emitting isotope <sup>125</sup>I.

Detailed Description Text - DETX (128):

Yet another preferred embodiment of radioactive coating of the present invention is that which has an isotope layer comprising tungsten-188 (W-188 or .sup.188 W) Tungsten-188 undergoes beta decay to become rhenium-188 (Re-188 or .sup.188 Re). Rhenium-188 undergoes beta decay as well, but emits a much higher energy particle than in W-188 decay.

The W-188 has a much longer half-life than does Re-188, thus the W-188 almost continuously creates more Re-188. This process is known as "generator," and the generator isotopes are referred to together by the shorthand W/Re-188 to indicate the relationship between the species. Generators are attractive for use in radiation delivery devices because they combine the energy levels of a short half-life species with the durability of the long half-life species. It is a general rule that particle energy and half-life are inversely proportional, and that long half-life species are more economical and practical to work with than short half-life species.

Detailed Description Text - DETX (132):

<u>Combinations of various isotopes provide another preferred embodiment</u> in that, <u>for example</u>, beta-emitting isotopes may be combined with gamma-emitting isotopes where gamma isotopes can deliver dosage to greater depths.

US-PAT-NO: 6416457

DOCUMENT-IDENTIFIER: US 6416457 B1

TITLE: System and method for intravascular ionizing tandem radiation therapy

DATE-ISSUED: July 9, 2002

INVENTOR-INFORMATION:

NAME CITY STATE ZIP CODE COUNTRY Urick; Michael J. Rogers MN N/A N/A Verin; Vitali E. Geneva N/AN/A CH Popowski; Youri G. Geneva N/AN/AСН

#### Detailed Description Text - DETX (6):

Refer now to FIG. 3 which illustrates the radiation source wire 16 utilized in the system 10 illustrated in FIG. 1. Elongate source wire 16 includes two primary components, namely an elongate shaft 30 and a distally disposed radioactive source 32. Radioactive source 32 may approximate a line source as illustrated. The radiation source 32 includes a radioisotope emitting ionizing radiation such as beta or gamma radiation. Preferably, the radioactive source 32 comprises a radiation emitting isotope such as Sr/Y-90, P-32, Y-90, Ce/Pr-144, Ru/Rh-106, W/Re-188, Ir-192, I-125, or Pd-103. Radiopaque markers 34 and 36 may be disposed on either side of the radioactive source 32 to facilitate intravascular placement utilizing x-ray fluoroscopy. Elongate source wire 16 may comprise a wide variety of different designs incorporating an elongate shaft 30 and a distally disposed radioactive source 32. Preferably, the source wire 16 comprises the design disclosed in U.S. Pat. No. 5,728,042 to Schwager, which is hereby incorporated by reference.

US-PAT-NO: 6685618 DOCUMENT-IDENTIFIER: US 6685618 B2

TITLE: Method for delivering radiation to an intraluminal site in the body

DATE-ISSUED: February 3, 2004

INVENTOR-INFORMATION:

NAME CITY STATE ZIP CODE COUNTRY Tam; Lisa A. Lake Forest CA N/A N/A

Trauthen; Brett A. Newport Beach CA N/A N/A Detailed

----- KWIC -----

#### Description Text - DETX (62):

Yet another preferred embodiment of thin film source of the present invention is that which has an isotope layer comprising tungsten-188 (W-188 or .sup.188 W). Tungsten-188 undergoes beta decay to become rhenium-188 (Re-188 or .sup.188 Re). Rhenium-188 undergoes beta decay as well, but emits a much higher energy particle than in W-188 decay. The W-188 has a much longer half-life than does Re-188, thus the W-188 almost continuously creates more Re-188. This process is known as "generator," and these generator isotopes are referred to together by the shorthand W/Re-188 to indicate the relationship between the species. Generators are attractive for use in radiation delivery devices because they combine the energy levels of a short half-life species with the durability of the long half-life species. It is a general rule that particle energy and half-life are inversely proportional, and that long half-life species are more economical and practical to work with than short half-life species.

### US 6685618 Claims 15 and 16:

- 14. A method of treating a site within a vessel as in claim 11, wherein said isotope is a gamma or beta emitting isotope.
- 15. A method as in claim 11, wherein the radioactive isotope comprises an isotope selected from the group consisting of P-32, I-125, Pd-103, W/Re-188, As-73, and Gd-153.

FYI: data at http://amdc.in2p3.fr/nubase/Nubase2003.pdf

# G. Audi et al. / Nuclear Physics A 729 (2003) 3–128

Nuclide	Mass exc (keV)	Mass excess (keV)		Excussion energy(keV)		Half-life			Ens	Reference	Decay modes and unensines (%)	
	72889.8 72794.6 72718.6	2.5 2.5 2.5	95.1737 171.180	0.0012 0.005		 #\$ #\$	1.0 0.4 1.4	3/2" (9/2*) (11/2")		I		
103 Pd 103 Pd 103 Pd w	-87479.1	2.9	784.79	0.10	16.991 25	d ns	0.019	5/2 <sup>+</sup> 11/2"	91 91	λ	ε=100 IT=100	
188W 188Re	-38667 -39016.1	3 1.4	) 13 <del>4</del> - ( 2	9.10	69.78 17.0040	d	0.05 0.0022	0+	02		β=100 β=100	

PGPUB-DOCUMENT-NUMBER: 20070272862 DOCUMENT-IDENTIFIER: US 20070272862 A1

TITLE: Method and Device for Remotely Communicating Using Photoluminescence or Thermoluminescence

PUBLICATION-DATE: November 29, 2007

INVENTOR-INFORMATION: Desbrandes, Robert; Van Gent, Daniel Lee

APPL-NO: 11/569357 DATE FILED: May 23, 2005

FOREIGN-APPL-PRIORITY-DATA:

COUNTRY APPL-NO DOC-ID APPL-DATE
FR 0405717 2004FR-0405717 May 26, 2004
FR 0503659 2005FR-0503659 April 12, 2005

PCT-DATA:

APPL-NO: PCT/EP05/52348 DATE-FILED: May 23, 2005 371-DATE: Nov 18, 2006

Brief Summary Text - BSTX (33):

[0047] Each group of entangled photons is made up of emitted photons together or at very short intervals by the same particle of the source, for example: electron, nucleus, atom, molecule.

The sources of ad hoc entangled photons usable for this invention are, for example:
[0048] Natural or artificial radioactive materials producing a radiation in a cascade.

For example, the Cobalt 60 atom emits almost simultaneously two gamma which are entangled and which can be used to irradiate a photoluminescent or thermoluminescent material. . . .

\_\_\_\_\_

#### FYI: data at http://amdc.in2p3.fr/nubase/Nubase2003.pdf

## G. Audi et al. / Nuclear Physics A 729 (2003) 3-128

Nuclide	Mass exc (keV)	Mass excess (keV)		Excitation energy(keV)		Half-life			Eas	Reference	Decay modes and intensities (%)
<sup>60</sup> Co <sup>60</sup> Co <sup>™</sup>	-61649.0 -61590.4	0.6 0.6	58.59	0.01	5.2713 10.467	y m	8,0008 8,006	5** 2*	00 00		β=100 IT:≈100; β=0.24 3

FYI "Production of Cobalt-60", HW-64073, W.L. Bunch, Hanford Atomic Products Operation, 2-29-1960, Page 2 http://www.osti.gov/bridge/servlets/purl/10152247-upHZQJ/native/10152247.pdf

Cobalt-60 is produced from cobalt-59 in isomeric states. The short half-life (10.4 minutes) isomer decays primarily by gamma emission to the long half-life (5.28 year) isomer which subsequently beta decays to an excited state of nickel 60. Photons of energy 1.17 and 1.33 MeV are emitted to stabilize the nickel 60 nucleus. For long irradiations the existence of the short half-life isomer can be neglected and the differential equation describing the formation of cobalt-60 is:

where N = number of atoms of cobalt (the subscripts refer to the isotopes),

 $\phi$  = conventional neutron flux, n/cm<sup>2</sup> sec,

= absorption cross section for 2200 m/sec neutrons,

= decay constant (= 0.693/7-) in sec<sup>-1</sup>,

t = exposure time in sec,

and the superscript refers to the original number of atoms present.

# Accelerated Emission of Gamma Rays from the 31-yr Isomer of <sup>178</sup>Hf Induced by X-Ray Irradiation

C. B. Collins, F. Davanloo, M. C. Iosif, R. Dussart,\* and J. M. Hicks Center for Quantum Electronics, University of Texas at Dallas, Richardson, Texas 75083

#### From Page 697:

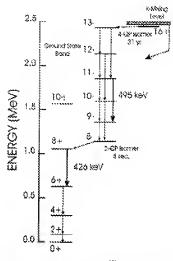


FIG. 3. Energy level diagram for <sup>116</sup>Hf showing a selection of levels relevant to this experiment. The spontaneous decay of the 51-yr, 167 isomer is shown by the thin arrows. The thick solid arrows show components studied in this work from the particular cascade from the decay forced by the x-ray irradiation. Heavy dotted arrows are inferred transitions needed to feed those shown.

During the spontaneous decay of the <sup>178</sup>Hf isomer the latter, being a member of the GSB, is fed only by cascade from the 4 sec, 8<sup>th</sup> bandhead.

\_\_\_\_\_

FYI: data at http://amdc.in2p3.fr/nubase/Nubase2003.pdf

G. Audi et al./Nuclear Physics A 729 (2003) 3-128

Nuclide Mass excess Excitation Half-life J\* Ens Reference Decay modes and (keV) energy(keV) intensities (%)

<sup>172</sup> Hf	-52444.3	2.1			STABLE		Ü <sup>+</sup>	94			IS=27.28 7
$^{178}\mathrm{Hf}^{sc}$	-51296.9	2.1	1147.423	0.005	4.0 s	0.2	8	94			IT=160
178 Hf*	-49998.6	2.1	2445.69	0.11	31 y	3	16+	94	94K1.A	$\mathbf{E}$	IT=100
178 H£P	-49870.8	2.2	2573.5	0.5	68 µs	2	$(14^{-})$	94			IT=100

97

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L128 ANSWER 6 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1997:616665 HCAPLUS
ΑN
     127:337716
DN
OREF 127:66223a,66226a
     Entered STN: 27 Sep 1997
ED
     Half-lives of isomeric levels of 87mSr, 111mCd, 113mIn
TΤ
     and 115mIn activated by 60Co \gamma-ray irradiation
     Nuclear Instruments & Methods in Physics Research, Section A: Accelerators,
SO
Spectrometers, Detectors, and Associated Equipment (1997), 397(2,3), 478-482
     CODEN: NIMAER; ISSN: 0168-9002
PB
     Elsevier
Journal Contents:
http://www.sciencedirect.com/science?_ob=PublicationURL&_tockey=%23TOC%235314%231997%239960299
97%2310802%23FLP%23&_cdi=5314&_pubType=J&_auth=y&_acct=C000055109&_version=1&_urlVersion=0&_us
erid=2502287&md5=ec893abac7bdb1577659de47a9e3985c
     Isomers having half-lives of a few hours were photoactivated by 60Co \gamma-ray irradiation
AB
     The .gamma .-rays emitted from the title isomers were measured with a low-background Ge
     detector. The decay data were analyzed by both a least-squares method and maximum
     likelihood method and the suitability of the latter method is discussed. The obtained
     half-lives are: 87\text{mSr} 2.811 ± 0.027 h, 111mCd 48.30 ± 0.15 min, 113mIn 1.663 ± 0.011 h
     and 115mIn 4.483 \pm 0.005 h.
     Gamma ray
ΤТ
        (irradiation; half-lives of isomeric levels of 87mSr,
        111mCd, 113mIn and 115mIn activated by 60Co γ-ray irradiation)
ΙT
     Nuclear energy level
        (isomer; half-lives of isomeric levels of 87mSr,
        111mCd, 113mIn and 115mIn activated by 60Co \gamma-ray irradiation)
     14191-71-0, Indium 115, properties 14336-64-2,
ΙT
     Cadmium 111, properties
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)
        (half-lives of isomeric levels of 87mSr, 111mCd,
        113mIn and 115mIn activated by 60Co γ-ray irradiation)
RN
     14191-71-0 HCAPLUS
CN
     Indium, isotope of mass 115 (CA INDEX NAME)
115<sub>In</sub>
     14336-64-2 HCAPLUS
RN
     Cadmium, isotope of mass 111 (CA INDEX NAME)
CN
```

111cd

#### L128 ANSWER 15 OF 42 HCAPLUS COPYRIGHT ACS on STN 1987:109090 HCAPLUS AN 106:109090 DNOREF 106:17751a,17754a ED Entered STN: 05 Apr 1987 Half-lives of cadmium-111m, indium-113m, and indium-115m ΤI ΑU Nemeth, Z.; Lakosi, L.; Pavlicsek, I.; Veres, A. Inst. Isot., Hung. Acad. Sci., Budapest, H-1525, Hung. CS Applied Radiation and Isotopes (1987), 38(1), 63-4 SO CODEN: ARISEF; ISSN: 0883-2889 DT Journal LA English CC 70-1 (Nuclear Phenomena) AΒ Bremsstrahlung from 4.5-MeV e on a Pt target was used to produce nuclear isomers in 3 nuclides through $(\gamma, \gamma')$ excitation of higher lying states and their half-lives were measured. The obtained values are $48.54 \pm 0.05$ min for 111mCd, $1.660 \pm 0.005$ h for 113mIn, and $4.485 \pm 0.004$ h for **115mIn**. The intensity ratio of 497/336 keV $\gamma$ transitions in the decay of 115mIn was $1.04 \pm 0.05 + 10-3$ . The methods of anal. and the results are discussed. ΙT Gamma ray (scattering of, by cadmium-111 and indium isotopes, inelastic) ΤТ **14191-71-0**, Indium-115, properties **14336-64-2**, Cadmium-111, properties RL: PRP (Properties) (nuclear energy levels of, excitation of isomeric, by γ-ray scattering) RN 14191-71-0 HCAPLUS CNIndium, isotope of mass 115 (CA INDEX NAME) 115<sub>In</sub>

14336-64-2 HCAPLUS

Cadmium, isotope of mass 111 (CA INDEX NAME)

RN CN

111cd

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L128 ANSWER 7 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1997:499756 HCAPLUS
ΑN
     127:199354
DN
OREF 127:38487a,38490a
ED
     Entered STN: 07 Aug 1997
     Determination of the elemental distribution in a sample using
TΙ
     neutron-induced gamma-ray emission tomography
ΑIJ
     Spyrou, N. M.; Sharaf, J. M.; Rajeswaran, S.; Mesbahi, E.
     Dep. Phys., Univ. Surrey, Guildford, GU2 5XH, UK
CS
SO
     Journal of Radioanalytical and Nuclear Chemistry (1997), 217(2),
     237-241
     CODEN: JRNCDM; ISSN: 0236-5731
     Elsevier
PB
     Journal
DT
LA
     English
Journal Contents:
http://springerlink.metapress.com/content/x1rp7g350421/?p=38c9a261e0da4b58903406eb28bb3651&pi=156
     The factor that affect accurate, quant. results to be obtained by neutron-induced \gamma-ray
AB
     emission tomog. are stated. The technique, which is a combination of NAA with
     computerized \gamma-ray emission tomog., would be enhanced by the use of multiple detector
     assemblies, in geometrical configurations, which simultaneously record the \gamma-rays emitted
     and improve detection efficiency. Developments in the past few years in positron
     emission tomog. where scanners made of single scintillation block detectors, cut into
     smaller segments, to form individual crystal detector elements and packed in ring around
     the radioactive object, are discussed. The coincident detection efficiency for
     annihilation photons and cascade \gamma-rays for such systems are considered and the
     possibilities of carrying out NIGET with coincident γ-ray tomog. were explored while
     indicating some of the limitations.
ΙT
     Scintillation detectors
        (calculated detection efficiencies for coincident \gamma-rays for Bi
        germanate system with 16 block detectors)
ΤТ
     Neutron activation analysis
        (determination of the elemental distribution in sample using neutron-induced
        gamma-ray emission tomog.)
     10198-40-0, Cobalt-60, analysis
                                        14265-76-0, Hafnium-179, analysis
TΤ
     14336-64-2, Cadmium-111, analysis
     RL: ANT (Analyte); ANST (Analytical study)
        (calculated detection efficiencies for coincident \gamma-rays for Bi
        germanate system with 16 block detectors)
RN
     10198-40-0 HCAPLUS
CN
     Cobalt, isotope of mass 60 (CA INDEX NAME)
 60Co
RN
     14265-76-0 HCAPLUS
CN
     Hafnium, isotope of mass 179 (CA INDEX NAME)
179 H f
     14336-64-2 HCAPLUS
RN
     Cadmium, isotope of mass 111 (CA INDEX NAME)
CN
111cd
```

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L128 ANSWER 8 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1996:369263 HCAPLUS
ΑN
     125:125256
DN
OREF 125:23289a,23292a
ED
     Entered STN: 26 Jun 1996
ΤI
     Nuclide yields in (\gamma, \gamma')-, (.gamma
     .,n)- and (\gamma,p) reactions at Eymax = 13.3 MeV
     Kunqurov, F. R.; Muminov, T. M.; Mukhamedov, A. K.; Saidmuradov, Zh.;
ΑU
     Salikhbaev, U. S.; Safarov, A. N.
     Samarkand State University, Samarkand, Uzbekistan
CS
     Izvestiya Akademii Nauk, Seriya Fizicheskaya (1996), 60(1),
SO
     201-205
     CODEN: IRAFEO
РΒ
     Nauka
     Journal
DT
LA
     Russian
CC
     70-1 (Nuclear Phenomena)
AΒ
     Nuclide yields in photonuclear reactions were determined at the electron accelerator of
     Samarkand University. The results are planned to be used in activation anal.
ST
     nuclide yield photonuclear reaction electron accelerator
ΙT
     Gamma ray
        (yields of various nuclides in (\gamma, n)-, (\gamma
        ,p)- and (\gamma,\gamma') reactions on medium and
        heavy nuclei at gamma-ray maximum energy of 13.3 MeV)
     14336-64-2, Cadmium-111, processes
ΙT
     RL: PEP (Physical, engineering or chemical process); PROC (Process)
        (excitation of isomeric state of 111Cd in (
        \gamma, \gamma') inelastic scattering on)
     14336-64-2 HCAPLUS
RN
CN
     Cadmium, isotope of mass 111 (CA INDEX NAME)
111cd
     14191-71-0, Indium-115, reactions
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (yields of metastable 114In and 115In in gamma-ray
        bombardment of)
     14191-71-0 HCAPLUS
RN
     Indium, isotope of mass 115 (CA INDEX NAME)
CN
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115 Tn

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L128 ANSWER 26 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1979:211284 HCAPLUS
ΑN
     90:211284
DN
OREF 90:33515a,33518a
ΤI
     Excitation of nuclear isomers by \gamma rays from cobalt-60
     Watanabe, Yoshihisa; Mukoyama, Takeshi
ΑU
CS
     Inst. Chem. Res., Kyoto Univ., Kyoto, Japan
SO
     Bulletin of the Institute for Chemical Research, Kyoto University (
     1979), 57(1), 72-82
     CODEN: BICRAS; ISSN: 0023-6071
     Journal
DT
     English
LA
CC
     70-2 (Nuclear Phenomena)
     Long-lived isomeric states in stable nuclei 111Cd [ 14336-64-2] and 115In [14191-71-0]
AΒ
     were excited by the Compton-scattered \gamma-rays from 60Co. The flux of the scattered
     photons at resonance was calculated in the single-scattering approximation. By using the
     observed \gamma activities of 111mCd and 115mIn and the calculated photon flux, the integral
     cross sections for the isomer production by photoexcitation of 111Cd and 115In were
     evaluated to be (3.5 \pm 0.4) + 10-25 and (1.9 \pm 0.1) + 10-25 cm<sup>2</sup>-eV, resp.
ΙT
     Gamma ray
        (scattering of, by cadmium-111 and indium-115, isomer
        excitation in)
ΙT
     14191-71-0, properties
                              14336-64-2, properties
     RL: PRP (Properties)
        (nuclear energy levels of, excited by γ-ray
        scattering, isomeric)
RN
     14191-71-0 HCAPLUS
CN
     Indium, isotope of mass 115 (CA INDEX NAME)
```

115<sub>In</sub>

RN 14336-64-2 HCAPLUS CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111cd

**10/599,555** search, **12/4/09**, Page 21 of 72 L128 ANSWER 31 OF 42 HCAPLUS COPYRIGHT ACS on STN 1973:460156 HCAPLUS ΑN 79:60156 DNOREF 79:9657a,9660a TΙ Nuclear quadrupole interaction studies by perturbed angular correlations Haas, H.; Shirley, D. A. ΑU CS Lawrence Berkeley Lab., Univ. California, Berkeley, CA, USA SO Journal of Chemical Physics (1973), 58(8), 3339-55 CODEN: JCPSA6; ISSN: 0021-9606 DTJournal LA English CC 75-2 (Nuclear Phenomena) Section cross-reference(s): 73 AΒ A comprehensive study was made of the applicability of .gamma .-ray angular correlations to the determination of quadrupole interactions in metals and insulating solids. Dynamic effects were studied in solns. and gases. A total of  $14 \gamma$ -ray cascades were employed. Several nuclear spins were confirmed and the quadrupole moments of 10 excited nuclear states were determined or estimated from the data. Quadrupole coupling consts. were determined for excited states of the following nuclei in metallic host lattices of the same element: 44Sc, 99Ru, 111Cd, 117In, 187Re, 199Hg. Coupling consts. were also measured for the following isotope (lattice) combinations: 99Ru(Zn, Cd, Sn, Sb), 100Rh(Zn, Ru, Cu5Zn3, Pd2Al, PdPb2), 111Cd(In, Hg, Tl, CdSb, Cd3Ag, Zn, Ga, In, Sn, Sb, Bi, AuIn, InBi, In2Bi), 115In(Cd), 117In(Cd, Sn), 131I(Te), 181Ta(HfB2, HfSi2), 204Pb(Cd, In, Sn, As, Sb, Bi, Hg, Tl, PdPb2). Systematic variations of e2qQ with host-lattice structure were observed and host and solute properties were found to be separable to some extent for nontransition metals. The nuclei 111Cd, 115In, 117In, 199Hg, and 204Pb were used to determine a total of 50 quadrupole coupling consts. in insulators, including 20 with non-zero asymmetry parameters, which give oscillatory but aperiodic correlation functions. It was strikingly (and exhaustively) demonstrated that good detns. of quadrupole coupling consts. could be made following isomeric transitions (with no elemental transmutation) and  $\beta$ -decay (with elemental transmutation). However, in no case was it possible to derive a coupling constant from a  $\gamma$ -ray cascade preceded directly by e-capture decay, presumably because the sudden creation of a K-hole, and the Auger and "shake-off" events that follow, destroy the chemical integrity of the species under study. Relaxation times were determined for a number of liquid samples. Studies of dimethyl-111Cdm in various buffer gases showed that the spin memory was lost in 1 collision with heavy mols., but that light mols. required several collisions. ΙT (angular correlation of, quadrupole interactions and perturbed) Nuclear energy level IT (of atomic nuclei, quadrupole moments and spins of) 14191-71-0, properties RL: RCT (Reactant); RACT (Reactant or reagent)

ΙT

(quadrupole coupling of, in cadmium and cadmium insulating compds.)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115<sub>In</sub>

IT 14336-64-2, properties RL: RCT (Reactant); RACT (Reactant or reagent) (quadrupole interactions of) 14336-64-2 HCAPLUS RN CN Cadmium, isotope of mass 111 (CA INDEX NAME)

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L128 ANSWER 33 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1970:138716 HCAPLUS
ΑN
     72:138716
DN
OREF 72:24773a,24776a
     Determination of the amounts of elements in ore and products of their
TΙ
     reprocessing by excitation of metastable states
     according to a (\gamma, \gamma' -) reaction
ΑU
     Kodiri, S.; Starchik, L. P.
CS
     USSR
SO
     Zavodskaya Laboratoriya (1970), 36(2), 191-5
     CODEN: ZVDLAU; ISSN: 0321-4265
DT
LA
     Russian
CC
     76 (Nuclear Technology)
AΒ
     With the aid of the bremsstrahlung obtained from a 4.2-MeV electron linear accelerator
     (average output current of 50 \muA) metastable states of the following nuclides were
     excited: 77Se, 87Sr, 89Y, 103Rh, 107Ag, 109Ag, 111Cd, 113In, 115In, 117Sn, 135Ba, 137Ba,
     176Lu, 179Hf, 195Pt, 19 7Au, and 199Hg. Either a 0.95-mm Pt target or a 2-mm Pb target
     was used. Time of the irradiation was of 3-4 half-lives for each element. The activity
     induced was measured by a \gamma-ray spectrometer with 40 + 50 mm NaI(Tl) crystal. Based on
     photopeak area detns. the following sensitivity limits (in mg) were found: 77Sem 0.16,
     87Srm 0.3, 89Ym 3.4, 103Rhm 2.01, 107Agm 0.24, 111Cdm 0.08, 115Inm 0.024, 117Snm 80.0,
     135Bam 7.1, 137Bam 1.9, 176L um 0.96, 179Hfm 0.1, 195Ptm 0.2, 197Aum 0.1, 199Hgm 0.9.
ΙT
     Gamma rays, chemical and physical effects
        (radioactivation by)
ΤТ
     Analysis
        (radioactivation, by gamma rays)
ΤТ
     13981-59-4, analysis 14191-71-0, analysis
     14191-88-9, analysis
                            14265-76-0, analysis
     14336-64-2, analysis
     RL: ANT (Analyte); ANST (Analytical study)
        (determination of, by gamma-ray activation)
     13981-59-4 HCAPLUS
RN
CN
     Tin, isotope of mass 117 (CA INDEX NAME)
117<sub>Sn</sub>
     14191-71-0 HCAPLUS
RN
CN
     Indium, isotope of mass 115 (CA INDEX NAME)
115<sub>In</sub>
     14191-88-9 HCAPLUS
RN
CN
     Platinum, isotope of mass 195 (CA INDEX NAME)
195Pt
     14265-76-0 HCAPLUS
RN
CN
     Hafnium, isotope of mass 179 (CA INDEX NAME)
179<sub>Hf</sub>
     14336-64-2 HCAPLUS
RN
CN
     Cadmium, isotope of mass 111 (CA INDEX NAME)
```

111cd

L128 ANSWER 40 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1951:46442 HCAPLUS

DN 45:46442

OREF 45:7890a-b

TI Natural radioactivity of indium

AU Cohen, S. G.

CS Hebrew Univ., Jerusalem

SO Nature (London, United Kingdom) (1951), 167, 779

CODEN: NATUAS; ISSN: 0028-0836

DT Journal

LA Unavailable

CC 3A (Nuclear Phenomena)

Natural radioactivity was sought in 2 pairs of neighboring isobars: In113-Cd113 and In115-Sn115. By use of a counter with an In cathode, some activity was found. The radiation could not be identified as K  $\mathbf{x}$  -rays of Cd; this fact indicates that any K-capture half life must be >1014 yrs.  $\beta$ -Rays from the In have an energy of a few hundred e.kv., as determined from absorption measurements. In115 probably decays by  $\beta$ -emission to Sn115, with a half life of .apprx.1014 yrs. By use of counters with different gas fillings, results were obtained which are consistent with the presence of a small amount of Cd L  $\mathbf{x}$ -rays, which may arise after decay of In113 by L-capture, but evidence is inconclusive owing to the small counting rates involved. With L-capture, a half life of .apprx.1012 yrs. is to be expected.

IT **14336-66-4**, Cadmium, isotope of mass 113

(from In115)

RN 14336-66-4 HCAPLUS

CN Cadmium, isotope of mass 113 (CA INDEX NAME)

113cd

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115<sub>In</sub>

```
L128 ANSWER 1 OF 42 HCAPLUS COPYRIGHT ACS on STN
     2005:636704 HCAPLUS
ΑN
DN
     144:75858
     Entered STN: 22 Jul 2005
ED
ΤI
     Investigation of formation of residual nuclei from natU by reactions with
     660 MeV protons
     Adam, J.; Katovsky, K.; Michel, R.; Balabekyan, A.; Tsoupko-Sitnikov, V.
ΑU
     M.; Kalinnikov, V. G.; Pronskikh, V. S.; Solnyshkin, A. A.; Stegailov, V.
     I.; Mashnik, S. G.; Prael, R. E.; Gudima, K. K.
     Joint Institute for Nuclear Research, Moscow, Russia
CS
     AIP Conference Proceedings (2005), 769(International Conference
SO
     on Nuclear Data for Science and Technology, 2005, Part 1), 1043-1046
     CODEN: APCPCS; ISSN: 0094-243X
     American Institute of Physics
PB
DT
     Journal
     English
LA
CC
     70-1 (Nuclear Phenomena)
     Thin natural uranium targets were irradiated by a 660 MeV proton beam from the Phasotron
AB
     accelerator at the Joint Institute for Nuclear Research in Dubna, Russia. Cross-sections
     of the formation of residual nuclei natU(p,xpyn)AZ Res are determined by methods of gamma
     spectroscopy. Until now, 43 long-lived (T1/2 > 100 days) isotopes were observed and
     their cross-sections determined More than 350 intermediate-lived (1 day < T1/2 <100
     days) and short-lived (T1/2 < 1 day) isotopes have been identified in the \gamma-spectra and
     many unanalyzed lines yet remain. Final results for 43 long-lived isotopes and upper
     cross-section limits for 27 long -lived isotopes are presented in this paper and compared
     with results by five different models.
     Nuclear reaction fragments
ΤT
        (formation of residual nuclei from natU by reactions with 660 MeV protons)
     10198-40-0P, Cobalt, isotope of mass 60, preparation
TΤ
     14265-77-1P, Hafnium, isotope of mass 178, preparation
     14314-35-3P, Tin, isotope of mass 119, preparation
     14336-66-4P, Cadmium, isotope of mass 113, preparation
     378782-82-2P, Niobium, isotope of mass 93m(16.1 \text{ yr}), preparation
     RL: FMU (Formation, unclassified); PNU (Preparation, unclassified); FORM
     (Formation, nonpreparative); PREP (Preparation)
        (formation of residual nuclei from natU by reactions with 660 MeV protons)
RN
     10198-40-0 HCAPLUS
CN
     Cobalt, isotope of mass 60 (CA INDEX NAME)
 60co
RN
     14265-77-1 HCAPLUS
CN
     Hafnium, isotope of mass 178 (CA INDEX NAME)
178<sub>Hf</sub>
RN
     14314-35-3 HCAPLUS
     Tin, isotope of mass 119 (CA INDEX NAME)
CN
119<sub>sn</sub>
RN
     14336-66-4 HCAPLUS
CN
     Cadmium, isotope of mass 113 (CA INDEX NAME)
113cd
RN
     378782-82-2 HCAPLUS
CN
     Niobium, isotope of mass 93m(16.1 yr) (CA INDEX NAME)
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Nb

#### 10/599,555 search, 12/4/09, Page 26 of 72 L128 ANSWER 2 OF 42 HCAPLUS COPYRIGHT ACS on STN 2004:328024 HCAPLUS ΑN DN141:92038 Entered STN: 22 Apr 2004 ED TΙ Devgaon (H3) chondrite: classification and complex cosmic ray exposure history Murty, S. V. S.; Rai, V. K.; Shukla, A. D.; Srinivasan, G.; Shukla, P. N.; ΑU Suthar, K. M.; Bhandari, N.; Bischoff, A. CS Planetary and Geosciences Division, Physical Research Laboratory, Ahmedabad, 380009, India Meteoritics & Planetary Science (2004), 39(3), 387-399 SO CODEN: MPSCFY; ISSN: 1086-9379 PΒ Meteoritical Society DT Journal LA English CC 53-9 (Mineralogical and Geological Chemistry) The Devgaon meteorite fell in India on Feb. 12, 2001 and was immediately collected. It AΒ is an ordinary chondrite having a number of SiO2-rich objects and some Ca, Al-rich inclusions. Olivines (Fa17-19) are fairly equilibrated, while pyroxenes (Fs4-20) are unequilibrated. Occasionally, shock veins are visible, but the bulk rock sample is very weakly shocked (S2). Chondrules and chondrule fragments are abundant. Based on chemical and petrol. features, Devgaon is classified as an H3.8 group chondrite. Several cosmogenic radionuclides ranging in half-lives from 5.6 d (52Mn) to 7.3 + 105 yr (26Al), noble gases (He, Ne, Ar, Kr, and Xe), and particle track d. were measured. The track d. in olivines from 5 spot samples varies between (4.6 to 9) + 106 cm-2 showing a small gradient within the meteorite. The light noble gases are dominated by cosmogenic and radiogenic components. Large amts. of trapped gases (Ar, Kr, and Xe) are present. In addition, $(n, \gamma)$ products from Br and I are found in Kr and Xe, resp. The average cosmic ray exposure age of $101 \pm 8$ Ma is derived based on cosmogenic 38Ar, 83Kr, and 126Xe. The track production rates correspond to shielding depths of .apprx.4.9 to 7.8 cm, indicating that the stone suffered type IV ablation. Low 60Co, high (22Ne/21Ne)c, and large neutron produced excesses at 80Kr, 82Kr, and 128Xe indicate a complex exposure history of the meteoroid. In the first stage, a meter-sized body was exposed for nearly 108 yr in the interplanetary space that broke up in .apprx.50 cm-sized fragments about a million years ago (stage 2), before it was captured by the Earth. ΙT Radionuclides, occurrence RL: GOC (Geological or astronomical occurrence); OCCU (Occurrence) (cosmogenic; classification and complex cosmic ray exposure history of Devgaon (H3) chondrite) ΙT **10198-40-0**, Cobalt-60, occurrence 13965-99-6, Xenon-129, occurrence 14683-11-5, Xenon-131, occurrence RL: GOC (Geological or astronomical occurrence); OCCU (Occurrence) (classification and complex cosmic ray exposure history of Devgaon (H3) chondrite) 10198-40-0 HCAPLUS RN Cobalt, isotope of mass 60 (CA INDEX NAME) CN 60Co

13965-99-6 HCAPLUS

14683-11-5 HCAPLUS

Xenon, isotope of mass 129 (CA INDEX NAME)

Xenon, isotope of mass 131 (CA INDEX NAME)

RN CN

RN CN

129xe

131<sub>Xe</sub>

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L128 ANSWER 3 OF 42 HCAPLUS COPYRIGHT ACS on STN
     2003:146929 HCAPLUS
ΑN
     138:176694
DN
     Entered STN: 27 Feb 2003
ED
     Differential effective cross sections of photonuclear reactions induced by
ТΤ
     bremsstrahlung of 11.8 MeV electrons in a thick tungsten target
ΑU
     Abibullaev, N. A.; Begimkulov, Kh. Kh.; Salikhbaev, U. S.
     Samark. Gos. Univ., Samarkand, Uzbekistan
CS
     O'zbekiston Fizika Jurnali (2002), 4(4), 277-282
SO
     CODEN: UJPZAR
PΒ
     Izdatel'stvo Fan
     Journal
DT
     Russian
LA
CC
     70-1 (Nuclear Phenomena)
AΒ
     In this work, the method of spherical activation spectrometry was used to measure the
     energy spectra and angular distribution of the bremsstrahlung from the MT-22S microtron
     electrons. Using the exptl. determined spectra of bremsstrahlung radiation with \mathrm{E}\gamma(\mathrm{max})
     = 11.8 MeV and the data on excitation functions, the effective cross sections for 63
     photonuclear reactions were studied.
ТТ
     Gamma ray interactions
        (differential effective cross section measurements of photonuclear
        reactions induced by bremsstrahlung of 11.8-MeV electrons in thick
        tungsten target)
ΤТ
     Nuclear reactions
        (photonuclear; differential effective cross section measurements of
        photonuclear reactions induced by bremsstrahlung of 11.8-MeV electrons
        in thick tungsten target)
ΙT
     14191-71-0, Indium-115, reactions
     RL: PEP (Physical, engineering or chemical process); PYP (Physical
     process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)
        (differential effective cross section measurements of gamma
        -ray inelastic scattering and neutron photoprodn. on 115In induced by
        bremsstrahlung of 11.8-MeV electrons in thick tungsten target)
     14191-71-0 HCAPLUS
RN
     Indium, isotope of mass 115 (CA INDEX NAME)
CN
115<sub>In</sub>
ΤТ
     14336-64-2, Cadmium-111, processes
     RL: PEP (Physical, engineering or chemical process); PYP (Physical
     process); PROC (Process)
        (differential effective cross section measurements of gamma
        -ray inelastic scattering on 111Cd with metastable state
        excitation induced by bremsstrahlung of 11.8-MeV electrons in
        thick tungsten target)
     14336-64-2 HCAPLUS
RN
     Cadmium, isotope of mass 111 (CA INDEX NAME)
CN
111cd
     14265-76-0, Hafnium-179, processes
TT
     RL: PEP (Physical, engineering or chemical process); PYP (Physical
     process); PROC (Process)
        (differential effective cross section measurements of gamma
        -ray inelastic scattering on 179,180Hf with metastable state
        excitation induced by bremsstrahlung of 11.8-MeV electrons in
        thick tungsten target)
RN
     14265-76-0 HCAPLUS
CN
     Hafnium, isotope of mass 179 (CA INDEX NAME)
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179<sub>Hf</sub>

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L128 ANSWER 4 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1999:561975 HCAPLUS
ΑN
     131:234316
DN
ED
     Entered STN: 03 Sep 1999
     Evaluation of a "best set" of average cross section measurements in the
TΙ
     235U(nth,f) neutron field
ΑU
     Mannhart, W.
CS
     Physikalisch-Technische Bundesanstalt Braunschweig, Braunschweig, Germany
SO
     Berichte des Forschungszentrums Juelich (1999), Juel-3660, 40-47
     CODEN: FJBEE5; ISSN: 0366-0885
DT
     Report
     English
LA
CC
     70-1 (Nuclear Phenomena)
AΒ
     Average cross sections measured in the thermal neutron-induced fission neutron field of
     235U were evaluated with a total of 200 data. The evaluation comprised the measurement
     of 30 different neutron reactions. The results were given and compared with 2 previous
     evaluations. X- ray and \gamma-ray emission probabilities of 169Yb and the half-life of 153Sm
     were also studied.
ΙT
     Gamma ray
       X-ray
        (x-ray \text{ and } \gamma-ray \text{ emission})
        probabilities of 169Yb)
ΙT
     15117-96-1, Uranium-235, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (average cross section measurements in thermal neutron-induced fission
        neutron field of)
     10198-40-0, Cobalt-60, formation (nonpreparative)
ΙT
     RL: FMU (Formation, unclassified); FORM (Formation, nonpreparative)
        (average cross section measurements in thermal neutron-induced fission
        neutron field of 235U)
     10198-40-0 HCAPLUS
RN
CN
     Cobalt, isotope of mass 60 (CA INDEX NAME)
60<sub>Co</sub>
ΤТ
     14191-71-0, Indium-115, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (average cross section measurements in thermal neutron-induced fission
        neutron field of 235U)
     14191-71-0 HCAPLUS
RN
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Indium, isotope of mass 115 (CA INDEX NAME)

115 Tn

#### L128 ANSWER 5 OF 42 HCAPLUS COPYRIGHT ACS on STN 1999:131580 HCAPLUS ΑN 130:227082 DNEntered STN: 02 Mar 1999 ED ΤI Performance analysis for disposal of mixed low-level waste. II: Results ΑU Waters, Robert D.; Gruebel, Marilyn M. CS Sandia National Lab, Albuquerque, NM, 87185-1395, USA Journal of Environmental Engineering (Reston, Virginia) (1999), 125(3), 258-266 SO CODEN: JOEEDU; ISSN: 0733-9372 PΒ American Society of Civil Engineers AΒ Ten sites were evaluated as potential locations for disposal of mixed low-level radioactive waste using a simple methodol. Three environmental pathways (water, atmospheric, and inadvertent intruder) were analyzed using generic transport models that were modified for site-specific conditions. The results were summarized by grouping the 58 evaluated radionuclides according to their half-lives and environmental mobility and by their limiting pathway (i.e., the pathway providing the lowest permissible radionuclide concentration in disposed waste of the three evaluated pathways). The results indicate that all evaluated sites have the tech. capability for disposal of some radionuclides in the waste. For most radionuclides, the intruder scenarios were more important in determining permissible radionuclide concns. than the other pathways, particularly for arid sites. For humid sites, if the water pathway is not the most limiting and the permissible radionuclide concentration is high, a more sophisticated and rigorous anal. of the water pathway may not be warranted. However, if the permissible concentration is relatively low, more refined analyses may produce higher permissible concns. based on addnl. site characterization data. ΤТ Air pollution Low-level radioactive wastes Simulation and Modeling, physicochemical (performance anal, for disposal of mixed low-level radioactive waste) ΙT 10198-40-0, Cobalt-60, occurrence 14336-66-4, Cadmium-113, occurrence 15726-30-4, Cesium-135, occurrence RL: POL (Pollutant); OCCU (Occurrence) (performance anal, for disposal of mixed low-level radioactive waste) RN10198-40-0 HCAPLUS Cobalt, isotope of mass 60 (CA INDEX NAME) CN

60co

113cd

135<sub>Cs</sub>

RN CN

RN CN 14336-66-4 HCAPLUS

15726-30-4 HCAPLUS

Cadmium, isotope of mass 113 (CA INDEX NAME)

Cesium, isotope of mass 135 (CA INDEX NAME)

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L128 ANSWER 9 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1995:601139 HCAPLUS
ΑN
     123:67459
DN
OREF 123:11869a,11872a
     Entered STN: 09 Jun 1995
ED
     Photoexcitation of metastable states in 111Cd, 113In and 115In in the 5-30
ΤТ
     MeV energy range
ΑU
     Safar, J.; Lakosi, L.; Veres, A.; Sekine, T.; Yoshihara, K.
CS
     Institute Isotopes, Hungarian Academy Sciences, Budapest, H-1525, Hung.
SO
     Proc. Int. Symp. Capture Gamma-Ray Spectrosc. Relat. Top., 8th (
     1994), Meeting Date 1993, 629-31. Editor(s): Kern, Jean.
     Publisher: World Sci., Singapore, Singapore.
     CODEN: 60VEAP
DT
     Conference
LA
     English
CC
     70-1 (Nuclear Phenomena)
AΒ
     The cross section of 111Cd(\gamma, \gamma')111mCd, 113In(\gamma, \gamma')113mIn and 115In(.gamma ., \gamma')115mIn
     reactions has been studied in the 5-30 MeV gamma-energy region. Excitation functions
      have been calculated in the framework of a statistical γ-ray cascade model considering
      open nucleon emission channels and including preequil. contribution. At around the
      neutron separation energy, the calculated cross sections show reasonable agreement with
      the majority of exptl. data available in the literature. Above the (\gamma, n) threshold the
      integral cross section calculated for 115In agrees well with our recently measured value
      within the exptl. errors. An exptl. value of 8.3 \pm 1.5 \text{ mb} MeV has been found for the
      113In(\gamma, .gamma .')113mIn reaction at 15 MeV endpoint energy also in agreement with our
     calcns.
ΙΤ
     Gamma ray
        (photoexcitation of metastable states in 111Cd, 113In and 115In)
ΙT
     14191-71-0, Indium 115, properties 14336-64-2,
     Cadmium 111, properties
     RL: PEP (Physical, engineering or chemical process); PRP (Properties);
     PROC (Process)
        (photoexcitation of metastable states in)
     14191-71-0 HCAPLUS
RN
     Indium, isotope of mass 115 (CA INDEX NAME)
CN
115<sub>In</sub>
```

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111cd

**10/599,555** search, **12/4/09**, Page 31 of 72 L128 ANSWER 10 OF 42 HCAPLUS COPYRIGHT ACS on STN 1993:416233 HCAPLUS AN DN119:16233 OREF 119:2941a,2944a Entered STN: 10 Jul 1993 ED ΤI Nuclear isomer excitation cross section in  $(\gamma, ...)$  $\operatorname{gamma.'})$  m reactions at 4-15 MeV ΑU Mazur, V. M.; Sokolyuk, I. V.; Bigan, Z. M.; Kobal, I. Yu. CS Inst. Yad. Issled., Ukraine SO Yadernaya Fizika (1993), 56(1), 20-5 CODEN: IDFZA7; ISSN: 0044-0027 DTJournal LA Russian CC70-1 (Nuclear Phenomena) Cross sections of  $\gamma$  inelastic scattering with **excitation** of isomer **states**  $\sigma m$  of 77Se, AΒ 79Br, 87Sr, 111Cd, 115In, and 137Ba nuclei were measured at 4-15-MeV energies in  $\Delta E = 0.5$ MeV steps. For  $(\gamma, \gamma')$ m reactions below the  $(\gamma, n)$  thresholds the energy dependence of the isomer ratios  $\eta = \sigma m/\sigma tot$  was considered. For nuclei whose spins of the ground and isomer states differ by  $\Delta J$  = 4, the dependence  $\eta$  = f(A) is studied in a wide mass interval 77 < A < 197. Increase of  $\eta$  = f(A) with increase of A is in qual. agreement with the statistical model predictions. ΙT Gamma ray (nuclear isomer state excitation by inelastic scattering of) 14191-71-0, Indium-115, properties 14336-64-2, ΙT Cadmium-111, properties RL: PRP (Properties) (nuclear isomer state excitation in, by γ-ray inelastic scattering)

115In

14191-71-0 HCAPLUS

RN

CN

RN 14336-64-2 HCAPLUS CN Cadmium, isotope of mass 111 (CA INDEX NAME)

Indium, isotope of mass 115 (CA INDEX NAME)

111cd

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L128 ANSWER 11 OF 42 HCAPLUS COPYRIGHT ACS on STN
    1990:162388 HCAPLUS
ΑN
     112:162388
DN
OREF 112:27387a,27390a
ED
     Entered STN: 28 Apr 1990
ΤI
     The Torino, H6, meteorite shower
ΑU
     Bhandari, N.; Bonino, G.; Callegari, E.; Castagnoli, G. Cini; Mathew, K.
     J.; Padia, J. T.; Queirazza, G.
     Ist. Cosmogeofis., Univ. Torino, Turin, Italy
CS
SO
     Meteoritics (1989), 24(1), 29-34
     CODEN: MERTAW; ISSN: 0026-1114
     Journal
DT
     English
LA
CC
     53-9 (Mineralogical and Geological Chemistry)
AΒ
     A meteorite shower fell at Torino, Italy, on 18 May, 1988. Petrog. studies indicate that
     the stone is an H6 chondrite having features of moderate to severe shock. Chemical
     analyses of the meteorite are reported. Cosmic ray-produced 3He, 21Ne, and 126Xe yield an
     exposure age of .apprx.48 Myr. The cosmic ray track densities in 3 fragments range
     between 1.8 + 105 to 5 + 105/\text{cm}2 suggesting .apprx.99% mass ablation in the atmospheric
     Twelve radioisotopes with half lives ranging between 5.6 days to 7.3 + 105 yr were
     measured with high precision (2-10%). Marginal signals were observed for several short-
     lived nuclides and upper limits were obtained for the activity levels of 8 radionuclides
     (24Na, 48Cr, 57Ni, 47Sc, 47Ca, 59Fe, 42Ar, and 44Ti), some of which have not been
     hitherto detected in fresh falls. The data are generally consistent with the nuclide
     production by galactic cosmic rays when modulation due to the solar cycle is taken into
     consideration. The preatm. radius of the chondrite was .apprx.20 cm, consistent with
     track densities and activity levels of 60Co, 26Al, and other radionuclides.
ΙT
     Trace elements, occurrence
     RL: OCCU (Occurrence)
        (in chondrite meteorite, of Torino, Italy)
     Fission fragments and products
     RL: PRP (Properties)
        (tracks of, d. of, in chondrite meteorite, ablation in atmospheric in relation
        to, of Torino, Italy)
     10198-40-0, Cobalt-60, occurrence
                                         13965-99-6, Xenon-129, occurrence
ΙT
     14683-11-5, Xenon-131, occurrence
     RL: OCCU (Occurrence)
        (in chondrite meteorite, of Torino, Italy)
RN
     10198-40-0 HCAPLUS
CN
     Cobalt, isotope of mass 60 (CA INDEX NAME)
60 Co
     13965-99-6 HCAPLUS
RN
CN
     Xenon, isotope of mass 129 (CA INDEX NAME)
129 Xe
     14683-11-5 HCAPLUS
RN
     Xenon, isotope of mass 131 (CA INDEX NAME)
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131<sub>Xe</sub>

#### L128 ANSWER 12 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1989:453313 HCAPLUS

DN 111:53313

OREF 111:8993a,8996a

- ED Entered STN: 20 Aug 1989
- TI Dose conversion factors for air, water, soil and building materials
- AU Holford, R. M.
- CS Chalk River Nucl. Lab., Chalk River, ON, KOJ 1J0, Can.
- SO At. Energy Can. Ltd., [Rep.] AECL (1988), AECL-9825, 449 pp. CODEN: AECRAN; ISSN: 0067-0367
- DT Report
- LA English
- CC 8-1 (Radiation Biochemistry)
  Section cross-reference(s): 59, 71
- This report provides dose conversion factors (DCFs) for 496 isotopes in 7 exposure situations. The tables include values for gamma dose at the body surface, beta dose at the body surface, beta dose to skin, DCFs for 24 individual organs, and an estimate of the DCF for the ED equivalent calculated according to ICRP rules. These calcns. were carried out in support of the Canadian Nuclear Waste-management Program, and are intended to form part of the assessment of the likely dose to humans from long-lived nuclides escaping from a disposal site.
- IT 10198-40-0, biological studies

RL: BIOL (Biological study)

(dosimetry of metastable and, in organs from building materials and environmental contamination)

RN 10198-40-0 HCAPLUS

CN Cobalt, isotope of mass 60 (CA INDEX NAME)

60<sub>Co</sub>

IT 14191-71-0, IN-115 element, biological studies

RL: ANT (Analyte); ANST (Analytical study)

(dosimetry of metastable and, in organs, from building materials and environmental contamination)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115<sub>In</sub>

#### 10/599,555 search, 12/4/09, Page 34 of 72 L128 ANSWER 13 OF 42 HCAPLUS COPYRIGHT ACS on STN 1989:413874 HCAPLUS ΑN 111:13874 DNOREF 111:2385a,2388a Entered STN: 08 Jul 1989 EDΤI Nuclear activation techniques for measuring direct and backscattered components of intense bremsstrahlung pulses ΑIJ Anderson, J. A.; Carroll, J. M.; Taylor, K. N.; Carroll, J. J.; Byrd, M. J.; Sinor, T. W.; Collins, C. B.; Agee, F. J.; Davis, D.; et al. Cent. Quantum Electron., Univ. Texas, Dallas, Richardson, TX, USA CS SO Nuclear Instruments & Methods in Physics Research, Section B: Beam Interactions with Materials and Atoms (1989), Volume Date 1988, B40-B41(2), 1189-92 CODEN: NIMBEU; ISSN: 0168-583X DT Journal LA English CC 71-1 (Nuclear Technology) High-voltage e accelerators used for bremsstrahlung generation can produce intense pulses of radiation with different endpoint energies. The energy spectrum can be changed by

AB High-voltage e accelerators used for bremsstrahlung generation can produce intense pulses of radiation with different endpoint energies. The energy spectrum can be changed by varying the charging voltage or by softening the photons with Compton scattering in a low atomic number material. The high dose rate and the flexible spectrum capabilities of the Aurora accelerator were used to investigate the potential for measuring the bremsstrahlung spectrum by photoactivation of nuclear isomeric states. Recent success in calibrating lower intensity sources has shown that gram-sized targets of isotopes, such as 115In, can be used to sample the incident x- rays at several discrete gateway energies. When irradiated at these energies the targets are excited to metastable states with lifetimes suitable for conventional counting after the flash.

IT Bremsstrahlung

(nuclear activation techniques for measuring direct and backscattered components of pulses of)

IT Accelerators

(electron, bremsstrahlung pulses from, nuclear activation techniques for measuring direct and backscattered components of)

IT 14336-64-2, Cadmium-111, properties

RL: PRP (Properties)

(isomeric activation of metastable, in measurement of direct and backscattered components of bremsstrahlung pulses)

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

111cd

115<sub>In</sub>

#### L128 ANSWER 14 OF 42 HCAPLUS COPYRIGHT ACS on STN 1987:563018 HCAPLUS ΑN DN107:163018 OREF 107:26081a,26084a Entered STN: 31 Oct 1987 ED ΤI Spin-flip magnetic quadrupole transitions in tin-115 and tin-117 ΑU Fukuda, M.; Nagai, Y.; Irie, T.; Ejiri, H. CS Dep. Phys., Osaka Univ., Toyonaka, 560, Japan Nuclear Physics A (1987), A470(1), 1-12 SO CODEN: NUPABL; ISSN: 0375-9474 DT Journal English LA Electromagnetic properties of low-lying excited states in 115Sn and 117Sn were studied by AΒ means of in-beam and off-beam e. gamma. spectroscopy. Magnetic quadrupole (M2) $\gamma$ matrix elements between the 1h11/2 and 1g7/2 quasi-n states in the Sn isotopes were reduced by factors qyeef/q.qamma .free = 0.21-0.29 over single quasi-particle values. The reduction rates are about the same as those for analogous quasi-p transitions. There is, however, a finite difference between the renormalization factor geff/gfree for the M2 $\gamma$ -decays and those for the analogous 1st-forbidden $\beta$ -decays. Electron internal conversion IΤ Gamma ray (from tin isotopes, from deuteron bombardment of indium-115 and decay of indium-117) **14191-71-0**, Indium-115, reactions ΤТ RL: RCT (Reactant); RACT (Reactant or reagent) (deuteron bombardment of, neutrons from) 14191-71-0 HCAPLUS RN Indium, isotope of mass 115 (CA INDEX NAME) CN115<sub>In</sub> ΙT 13981-59-4, Tin-117, properties RL: PRP (Properties) (nuclear energy levels of, from decay of indium-117, spin-flip magnetic quadrupole transitions of) RN 13981-59-4 HCAPLUS

Tin, isotope of mass 117 (CA INDEX NAME)

CN

117<sub>Sn</sub>

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L128 ANSWER 16 OF 42 HCAPLUS COPYRIGHT ACS on STN
    1986:77103 HCAPLUS
AN
     104:77103
DN
OREF 104:12143a,12146a
ED
     Entered STN: 08 Mar 1986
TΙ
     The strength of the spallation neutron flux of SINQ for radiation damage
     fusion technology
     Hegedus, F.; Green, W. V.; Stiller, P.; Green, S.; Herrnberger, V.;
ΑU
     Stiefel, U.; Victoria, M.
CS
     Swiss Fed. Inst. Reactor Res., Wuerenlingen, 5303, Switz.
SO
     EIR-Ber. (1985), 579, 40 pp.
     CODEN: EIBEDK; ISSN: 0531-6758
DT
     Report
LA
     English
CC
     71-2 (Nuclear Technology)
     Section cross-reference(s): 70
     The SINQ spallation n source was simulated and its source strength measured in terms of
AB
     the radiation damage parameters: total flux of spallation n with energy E < 1.0 = MeV;
     spectral energy distribution; He and displacement damage production rates; and their
     ratio. This simulation used the p beam of the TRIUMF accelerator in Canada, and its
     molten Pb beam stop as the source of spallation n. The He production per p of beam was
     measured for several materials by vacuum extraction, inside a calibrated mass
     spectrograph; the spectrum, flux extraction inside a calibrated mass spectrograph; the
     spectrum, flux intensity, and the displacement radiation damage parameter were measured
     by multiple foil activation flux-unfolding combined with radiation damage calcns. The
     He/dpa (displacements per atom) ratio matches the fusion reactor 1st wall case; but the
     He production in Fe per mA-yr at a radial distance of 15 cm is estimated, to be 6 appm
     compared to 310 appm at the end of life in the 1st wall of NET.
     Radiation, chemical and physical effects
ΙT
        (damage by, of fusion reactor first wall materials)
ΙT
     Nuclear fusion reactors
        (first walls, radiation damage to, spallation neutron flux in relation
        to)
     10198-40-0P, preparation
ΙT
     RL: PREP (Preparation)
        (production of, by neutron bombardment of copper or nickel, gamma
        ray activities of product nuclei in relation to)
RN
     10198-40-0 HCAPLUS
CN
     Cobalt, isotope of mass 60 (CA INDEX NAME)
60 Co
     14191-71-0P, preparation
ΙT
     RL: PREP (Preparation)
        (production of, by neutron bombardment of indium, gamma ray
        activities of product nuclei from)
     14191-71-0 HCAPLUS
RN
     Indium, isotope of mass 115 (CA INDEX NAME)
CN
```

115<sub>In</sub>

L128 ANSWER 17 OF 42 HCAPLUS COPYRIGHT ACS on STN 1985:156656 HCAPLUS ΑN 102:156656 DNOREF 102:24529a,24532a ED Entered STN: 04 May 1985 Long-lived radionuclides in low-level waste ΤI ΑIJ Cline, James E.; Coe, Larry J. CS Sci. Appl., Inc., USA Proc. Annu. Participants' Inf. Meet.: Low-Level Waste Manage. Program SO (1983), Issue 8308106, DE84 004134, 404-116 Publisher: NTIS, Springfield, Va. CODEN: 53ITAD  $\mathsf{DT}$ Conference LA English CC 71-11 (Nuclear Technology) AΒ The radionuclide contents were studied of LWR-generated low-level waste. The objectives of the study are: (1) to analyze, using verified techniques, 150 archived samples for specified  $\beta$ - and x- ray-emitting nuclides that had not previously been analyzed; (2) to analyze 20 new samples obtained from operating plants for all relevant nuclides and compare them to previous data to ascertain trends; (3) to develop empirical scaling factors through the use of which concns. of hard-to-analyze nuclides can be estimated from analyses of the  $\gamma$ -ray emiting nuclides. The new samples are analyzed and the results are summarized and interpreted. Scaling factor development is discussed. Factors are presented that relate 63Ni and 59Ni to 60Co for PWRs and to 58Co for BWRs, 90Sr to 137Cs for BWRs and 241Pu, 239Pu, 241Am, and 244Cm to 144Ce for all LWRs. ST radioelement radioactive waste LWR; scaling factor radioelement radioactive waste Radioactive wastes TΤ (low-level, determination of long-lived radionuclides in, from LWR, scaling factors for) ΙT **10198-40-0**, analysis **15726-30-4**, analysis RL: ANT (Analyte); ANST (Analytical study) (determination of, in low-level radioactive wastes) RN 10198-40-0 HCAPLUS Cobalt, isotope of mass 60 (CA INDEX NAME) CN 60Co 15726-30-4 HCAPLUS RN

Cesium, isotope of mass 135 (CA INDEX NAME)

CN

135cs

#### 10/599,555 search, 12/4/09, Page 38 of 72 L128 ANSWER 18 OF 42 HCAPLUS COPYRIGHT ACS on STN 1982:188345 HCAPLUS ΑN DN96:188345 OREF 96:30941a,30944a Nuclear data for $\mathbf{x}-$ or gamma- $\mathbf{ray}$ spectrometer TΙ efficiency calibrations ΑU Hoppes, D. D.; Hutchinson, J. M. R.; Schima, F. J.; Unterweger, M. P. CS Natl. Bur. Stand., Washington, DC, USA NBS Special Publication (United States) (1982), 626, 85-99 SO CODEN: XNBSAV; ISSN: 0083-1883 DT Journal English LA CC 70-1 (Nuclear Phenomena) Section cross-reference(s): 71, 73 Half-lives measured with reference ionization chambers are reported for 22Na, 24Na, 46Sc, AB 51Cr, 54Mn, 57Co, 58Co, 59Fe, 60Co, 65Zn, 67Ga, 75Se, 85Sr, 88Y, 99Mo, 99mTc (two chemical forms), 109Cd, 110mAq, 111In, 113Sn, 123I, 125I, 125Sb, 127Xe, 131I, 131mXe, 133Xe, 133Ba, 134Cs, 137Cs, 139Ce, 140Ba, 140La, 141Ce, 144Ce, 152Eu, 153Gd, 154Eu, 155Eu, 169Yb, 181W, 192Ir, 195Au, 198Au, 201Tl, 203Hg, 203Pb, 207Bi, and 228Th. γ-Ray probabilities per decay measured with calibrated Ge spectrometry systems are given for 88Y, 99Mo, 125Sb, 131I, 154Eu, and 228Th. The $\gamma$ -ray to K-x-ray emission-rate ratio for 109Cd and the $\gamma$ -ray probability per decay for 241Am as measured with NaI(Tl) systems of known efficiency are given. ΙT X-ray spectrometry (data for calibrations in) ΤТ Gamma ray (spectrometry, data for calibrations in) ΙT Nuclear spectrometers (gamma-ray, efficiency calibration of germanium) ΙT **14683-11-5**, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (decay of metastable, half-life and γ-rays from) RN 14683-11-5 HCAPLUS Xenon, isotope of mass 131 (CA INDEX NAME) CN 131xe

ΙT

RN

CN

60co

10198-40-0, reactions

10198-40-0 HCAPLUS

RL: RCT (Reactant); RACT (Reactant or reagent) (decay of, half-life and γ-rays from)

Cobalt, isotope of mass 60 (CA INDEX NAME)

```
L128 ANSWER 19 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1982:76244 HCAPLUS
ΑN
     96:76244
DN
OREF 96:12427a,12430a
     Decay calculations on medium-level and actinide-containing wastes from the
ТΤ
     LWR fuel cycle. Part I. Basic data evaluation including activity and
     thermal decay power
ΑU
     Haug, H. O.
CS
     Inst. Heisse Chem., Kernforschungszent. Karlsruhe, Karlsruhe, Fed. Rep.
     Ger.
SO
     Kernforschungszent. Karlsruhe, [Ber.] KFK (1981), KFK 3221, 71 pp.
     CODEN: KKBRAY; ISSN: 0303-4003
     Report
DT
     German
LA
CC
     71-11 (Nuclear Technology)
     Section cross-reference(s): 55, 60
AΒ
     A number of basic data on medium-level and actinide-containing waste streams from the LWR
     fuel cycle were evaluated, and the activity and thermal decay power were calculated for
     the nuclide inventories of cladding hulls and fuel assembly structural materials, for
     feed clarification sludge, medium-level aqueous process waste, low-level solid
     transuranium waste and for medium-level reactor operating waste. The activity as a
     function of decay time of the medium-level wastes decreases within 500 to 600 yr by 1-3
     orders of magnitude and is at the same time .apprx.1-2 orders of magnitude lower than the
     activity of the high-level waste. The thermal decay power of the medium-level wastes
     decreases after 10-100 yr by .apprx.3 orders of magnitude and is about a factor of 10-100
     less than that of high-level waste. In the very long term, the residual activity (and
     thermal power) decreases only slowly due to the long half-lives of the dominant
     actinides. The activity after >1000 yr is .apprx. 1-2 orders of magnitude lower than
     that of high-level waste, the low-level transuranium waste by a factor 10 to 4, resp.
     The activity per unit volume of the packaged waste of the medium-level and actinide-
     containing wastes, owing to the bigger volume of the conditioned wastes, is lower by 2-4
     orders of magnitude up to .apprx.500 \text{ yr}. After >1000 \text{ yr} the activities per unit volume
     are lower by a factor of 20-200 than that of high-level waste.
ΙT
     Radioactive wastes
        (actinide-containing and medium-level, from LWR fuel cycle, decay calcns.
        on)
ΙT
     Actinides
     Fission fragments and products
     RL: PRP (Properties)
        (activity of, in aqueous medium-level radioactive wastes)
ΙT
     Nuclear reactor fuels and fuel elements
        (cycles, plutonium estimated losses in LWR)
TT
     14314-35-3, properties
                              14390-73-9, properties
     RL: PRP (Properties)
        (nuclear decay calcns. on metastable, in radioactive waste)
RN
     14314-35-3 HCAPLUS
CN
     Tin, isotope of mass 119 (CA INDEX NAME)
119<sub>Sn</sub>
     14390-73-9 HCAPLUS
CN
     Tellurium, isotope of mass 125 (CA INDEX NAME)
125те
     10198-40-0, properties
ΤТ
     RL: PRP (Properties)
```

(nuclear decay calcns. on radioactive wastes containing)

Cobalt, isotope of mass 60 (CA INDEX NAME)

10198-40-0 HCAPLUS

RN

CN

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L128 ANSWER 20 OF 42 HCAPLUS COPYRIGHT ACS on STN
AN
    1981:522261 HCAPLUS
     95:122261
DN
OREF 95:20397a,20400a
TΙ
    Beta decay rates for s-process studies
ΑU
     Cosner, Kenneth; Truran, James W.
CS
     Dep. Astron., Univ. Illinois, Urbana, IL, USA
     Astrophysics and Space Science (1981), 78(1), 85-94
SO
     CODEN: APSSBE; ISSN: 0004-640X
DT
     Journal
LA
     English
CC
     70-5 (Nuclear Phenomena)
AB
     The rates for a variety of \beta decay processes were determined as a function of temperature
     for nuclei which can participate in the s-process production of heavy elements, occurring
     in the presence of the 22Ne(\alpha,n)25Mg n source operating in the convective He shells of
     thermally pulsing stars. Specifically, calculated half-lives are presented for e-
     emission, e+ emission, and e- capture over the temperature range 108-109 K.
ΙT
     Stars
        (nucleosynthesis in, slow neutron capture, rate calcns. for \beta
        decay processes as function of temperature in relation to)
ΙT
     Atomic nuclei
        (nucleosynthesis of, rate calcns. for \boldsymbol{\beta} decay processes as
        function of temperature in relation to s-process)
ΙT
     Beta decay
     Electron-capture decay
        (rate calcns. for, as function of temperature, s-process nucleosynthesis in
        relation to)
     10198-40-0, reactions 14191-71-0, reactions
ΙT
     RL: PRP (Properties)
        (beta decay half-life of, temperature dependence of, s-process
        nucleosynthesis in relation to)
     10198-40-0 HCAPLUS
RN
CN
     Cobalt, isotope of mass 60 (CA INDEX NAME)
60co
RN
     14191-71-0 HCAPLUS
CN
     Indium, isotope of mass 115 (CA INDEX NAME)
```

115<sub>In</sub>

L128 ANSWER 21 OF 42 HCAPLUS COPYRIGHT ACS on STN 1981:146578 HCAPLUS ΑN 94:146578 DNOREF 94:23893a,23896a High resolution  $\gamma$  spectra of 40-44 MeV photon activation products. Part 3. A summary of  $\gamma$  rays, radionuclides and nuclear interferences observed Williams, D. R.; Hislop, J. S. ΑU Environ. Med. Sci. Div., AERE, Harwell, UK CS U. K. At. Energy Res. Establ., Rep. (1980), AERE-R 9022, 32 pp. SO CODEN: UKRGAL; ISSN: 0436-9734 Report DT LA English 70-2 (Nuclear Phenomena) CC Section cross-reference(s): 79 A table of  $\gamma$ -rays observed in the high resolution .gamma . ray spectra of 40-44 MeV  $\gamma$ AΒ photon activation products is presented. This table is arranged in order of increasing .gamma .-ray energy and the parent isotopes, their half-lives and their inactive precursors are identified. Nuclear interferences caused by production of an active isotope from different parent elements have been identified and evaluated quant. These are also tabulated. ST activation analysis gamma table Gamma ray ΙT (tables of) ΙT Radiochemical analysis (activation, gamma-ray tables for) 10198-40-0, uses and miscellaneous 14191-71-0, uses IΤ and miscellaneous RL: USES (Uses) (nuclear interference of, in activation anal.) RN 10198-40-0 HCAPLUS CN Cobalt, isotope of mass 60 (CA INDEX NAME)

14191-71-0 HCAPLUS RN

CN Indium, isotope of mass 115 (CA INDEX NAME)

115<sub>In</sub>

60Co

### **10/599,555** search, **12/4/09**, Page 42 of 72 L128 ANSWER 22 OF 42 HCAPLUS COPYRIGHT ACS on STN 1980:223010 HCAPLUS ΑN 92:223010 DNOREF 92:35971a,35974a Dose-rate conversion factors for external exposure to photon and electron TΙ radiation from radionuclides occurring in routine releases from nuclear fuel cycle facilities ΑIJ Kocher, D. C. Health Saf. Res. Div., Oak Ridge Natl. Lab., Oak Ridge, TN, 37830, USA CS Health Physics (1980), 38(4), 543-621 SO CODEN: HLTPAO; ISSN: 0017-9078 Journal DT English LA CC 71-9 (Nuclear Technology) Section cross-reference(s): 8 Dose-rate conversion factors for external exposure to photon and e radiation were AΒ calculated for 240 radionuclides of potential importance in routine releases from nuclear fuel cycle facilities. Exposure modes considered are immersion in contaminated air, immersion in contaminated water, and irradiation from a contaminated ground surface. each exposure mode, dose-rate conversion factors for photons and e were calculated for tissue-equivalent material at the body surface of an exposed individual. Dose-rate conversion factors for photons only were calculated for 22 body organs. Actinides ΤТ Fission fragments and products Radioelements, biological studies RL: BIOL (Biological study) (dose rate at body surface and dose-conversion factors for various organs exposed to) ΙT Nuclear reactor fuel reprocessing (dose-rate conversion factors for external exposure to radiation from radioelements released in) ΙT Gamma ray (dose-rate conversion factors for, for various radioelements) Dosimetry (dose-rate conversion factors, for radioelements occurring in routine releases from fuel reprocessing facilities) 10198-40-0, biological studies 14191-71-0, biological studies

ΙT

ΙT

RL: BIOL (Biological study)

(dose rate at body surface and dose-conversion factors for various organs exposed to)

RN 10198-40-0 HCAPLUS

CN Cobalt, isotope of mass 60 (CA INDEX NAME)

60co

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

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L128 ANSWER 23 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1980:157197 HCAPLUS
ΑN
     92:157197
DN
OREF 92:25355a,25358a
     A consistent set of nuclear-parameter values for absolute INAA
TΤ
ΑU
     Heft, Robert E.
SO
     DOE Symposium Series (1979), Volume Date 1978, 49 (Comput. Act.
     Anal. Gamma-Ray Spectrosc.), 495-510
     CODEN: DOESD6; ISSN: 0164-2022
AB
     Gamma spectral anal. of irradiated material can be used to determine absolute
     disintegration rates for specific radionuclides. These data, together with measured
     values for the thermal and epithermal neutron fluxes and irradiation, cooling, and
     counting-time values, are all the exptl. information required to do absolute instrumental
     neutron activation anal. (INAA). The calcns. required to go from product photon emission
     rate to target nuclide amount depend on values used for the thermal-neutron-capture cross
     section, the resonance absorption integral, the half- life, and photon branching ratios.
     Values for these parameters were determined by irradiating and analyzing a series of
     elemental stds. The results of these measurements were combined with values reported by
     other workers to arrive at a set of recommended values for the consts. Values for 114
     nuclides are listed.
     Radiochemical analysis
ΙT
        (neutron activation, instrumental, nuclear parameter values for absolute)
                             14191-71-0, properties 14191-88-9, properties
ΙT
     13981-59-4, properties
                                                                                14336-64-2,
     RL: PRP (Properties)
        (gamma rays from metastable, following neutron capture in
        activation anal., consts. for)
RN
     13981-59-4 HCAPLUS
CN
     Tin, isotope of mass 117 (CA INDEX NAME)
117<sub>Sn</sub>
RN
     14191-71-0 HCAPLUS
     Indium, isotope of mass 115 (CA INDEX NAME)
CN
115<sub>In</sub>
RN
     14191-88-9 HCAPLUS
CN
     Platinum, isotope of mass 195 (CA INDEX NAME)
     14336-64-2 HCAPLUS
RN
CN
     Cadmium, isotope of mass 111 (CA INDEX NAME)
111cd
TТ
     13967-67-4, properties 14191-71-0, properties 14191-88-9, properties 14265-76-0,
        (neutron cross sections of, for activation anal.)
RN
     13967-67-4 HCAPLUS
CN
     Iridium, isotope of mass 193 (CA INDEX NAME)
193<sub>Ir</sub>
     14191-71-0 HCAPLUS
RN
CN
     Indium, isotope of mass 115 (CA INDEX NAME)
115<sub>In</sub>
     14191-88-9 HCAPLUS
RN
     Platinum, isotope of mass 195 (CA INDEX NAME)
CN
195pt
     14265-76-0 HCAPLUS
RN
CN
     Hafnium, isotope of mass 179 (CA INDEX NAME)
```

179Hf

L128 ANSWER 24 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1980:48454 HCAPLUS

DN 92:48454

OREF 92:7927a,7930a

- TI **Excitation** of indium-115, indium-113, and cadmium-111 nuclei in the photo-free annihilation of positrons
- AU Vishnevskii, I. N.; Zheltonozhskii, V. A.; Svyato, V. P.; Trishin, V. V.
- CS Inst. Yad. Issled., Leningrad, USSR
- SO Izvestiya Akademii Nauk SSSR, Seriya Fizicheskaya (1979), 43(10), 2142-7

CODEN: IANFAY; ISSN: 0367-6765

- DT Journal
- LA Russian
- CC 70-2 (Nuclear Phenomena)
- Excitation of the isomeric levels of 336.2 keV in 115In [ 14191-71-0], 391.7 keV in 113In [14885-78-0], and 150.8 and 245.4 keV in 111Cd [14336-64-2] in photon-free e+ annihilation was studied. Natural In or Cd targets together with a planar 64Cu e+ sources were in a sandwich-type arrangement. The deexcitation . gamma.-rays of the above levels was measured with Ge(Li) detectors. The cross sections of the level excitation was determined Direct excitation of the above levels with e+ and excitation by  $\gamma$ -rays was excluded. The levels at 1077.8 keV in 115In and 1129.4 keV in 113In were excited by the photon-free positron annihilation and the levels at 336.2 and 391.7 keV, resp., were populated by the deexcitation of the higher levels. The level populating the states at 150.8 and 245.4 keV in 111Cd is not known. The cross sections agree with those of T. Mikoyama et al. (1972) but do not agree with theor. predictions of D. P. Grechukhin et al. (1977).
- IT Gamma ray

(from cadmium-111 and indium isotope level deexcitation)

IT 14191-71-0, properties 14336-64-2, properties

RL: PRP (Properties)

(nuclear energy level excitation of, by positron annihilation)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115<sub>In</sub>

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

#### L128 ANSWER 25 OF 42 HCAPLUS COPYRIGHT ACS on STN ΑN 1979:547062 HCAPLUS 91:147062 DNOREF 91:23615a,23618a TΙ The 852 keV first-forbidden beta transition in cadmium-115g ΑU Lakshminarayana, S.; Rao, M. Srinivasa; Rao, V. Seshagiri; Sastry, D. L. CS Lab. Nucl. Res., Andhra Univ., Waltair, India Nuovo Cimento della Societa Italiana di Fisica, A: Nuclei, Particles and SO Fields (1979), 52A(1), 92-8 CODEN: NIFAAM; ISSN: 0369-4097 DT Journal English LA CC 70-2 (Nuclear Phenomena) AΒ $\beta$ y Angular-correlation expts. were performed on the 852 keV $\beta$ -261 keV $\gamma$ cascade in the ground-state decay of 115Cd [14336-68-6] by using a conventional fast-slow coincidence set-up. The present integral-correlation experiment confirmed a 3/2- spin-parity assignment for the 597-keV level in 115In [14191-71-0], while the differentialcorrelation measurements revealed a small energy dependence of the correlation function. The nonunique 1st-forbidden 852-keV $\beta$ transition does not obey the $\xi$ -approximation ΙT Gamma rav (of indium-115, from decay of cadmium-115) 14336-68-6, reactions ΙT RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent)

(decay of, gamma-ray spectrum in)

(nuclear energy levels of, from cadmium-115 decay)

12587-47-2P

RL: PREP (Preparation)

14191-71-0, properties

RL: PRP (Properties)

(from cadmium-115 decay)

TΤ

IΤ

### L128 ANSWER 27 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1979:111154 HCAPLUS

DN 90:111154

OREF 90:17445a,17448a

- TI Nuclear excitation of indium-115 by positron annihilation with K-shell electrons
- AU Watanabe, Yoshihisa; Mukoyama, Takeshi; Shimizu, Sakae
- CS Inst. Chem. Res., Kyoto Univ., Kyoto, Japan
- SO Physical Review C: Nuclear Physics (1979), 19(1), 32-7 CODEN: PRVCAN; ISSN: 0556-2813
- DT Journal
- LA English
- CC 70-2 (Nuclear Phenomena)
- When a e+ annihilates with a K-shell e-, the excess energy liberated may be given to the nucleus involved. When this energy is just the right energy for any **excited state** within its level width of a fraction of an eV, nuclear excitation can be expected, and if the excited level **cascades** down to an isomeric state with appreciable lifetime, the occurrence of this annihilation mode can be confirmed by observing  $\gamma$ -transitions or conversion e- from this isomeric level. **Gamma-rays from 115mIn** [14191-71-0] were observed after irradiation of natural In foils by e+ from 64Cu. By using the observed induced  $\gamma$  activity of 115mIn and assuming this phenomenon to be a 2-step process, the cross sections of nuclear excitation by e+ annihilation for the 1078- and 1464-keV levels of 115In were evaluated as  $(3.9 \pm 1.4) + 10-24$  cm2 and  $(1.4 \pm 0.5) + 10-22$  cm2, resp.
- IT 12585-85-2
  - RL: RCT (Reactant); RACT (Reactant or reagent)
    (annihilation of, with K-shell electrons, cross sections for excitation of indium-115 levels by)
- IT 14191-71-0, properties
  - RL: PRP (Properties)

(nuclear energy levels of, excited by positron annihilation with K-shell electrons, cross sections for)

10/599,555 search, 12/4/09, Page 47 of 72 L128 ANSWER 28 OF 42 HCAPLUS COPYRIGHT ACS on STN 1976:96564 HCAPLUS ΑN DN84:96564 OREF 84:15679a,15682a Determination of the sign and magnitude of the nuclear quadrupole ТΤ interaction by  $\beta-\gamma$  directional correlations Raghavan, R. S.; Raghavan, P.; Kaufmann, E. N. ΑU Bell Lab., Murray Hill, NJ, USA CS Physical Review C: Nuclear Physics (1975), 12(6), 2022-32 SO CODEN: PRVCAN; ISSN: 0556-2813 DT Journal English LA CC 70-2 (Nuclear Phenomena) AΒ Expts. are reported which demonstrate that the sign and magnitude of the quadrupole interaction of excited nuclear states can be reliably determined by means of timedifferential  $\beta$ -.gamma . directional correlations with radioactive sources embedded in noncubic single crystals. The 828-keV level of 115In [14191-71-0] and the 247-keV level of 111Cd [14336-64-2], fed by an allowed and a unique first-forbidden  $\beta$ -decay, resp., were investigated by this method. The coupling consts. e2qQ/h of these levels in Cd metal at 298°K, are -146(5) MHz for 115In and +125(4) MHz for 111Cd. The pos. coupling constant for 111Cd in Cd coupled with a pos. quadrupole moment Q of this level implies a pos. elec. field gradient in Cd confirming recent theor. predictions. The neg. e2qQ for the 115In level in Cd metal correspondingly indicates a neg. Q for this level in conformity with its description as a member of a K = 1/2 rotational band. A discussion is included of the major concepts underlying the various methods available now for determination of the sign of e2qQ. Also a detailed account of theor. formulas necessary to evaluate allowed and forbidden eta-.gamma . perturbed angular correlations and a brief discussion of the significance of the technique for further studies of nuclear quadrupole

IT Gamma ray

(angular correlations of, with  $\boldsymbol{\beta}$  particles, nuclear quadrupole interactions from)

IT Nuclear energy level

(of cadmium-111 and indium-115, quadrupole interactions of)

IT 14191-71-0, properties

RL: PRP (Properties)

(nuclear energy levels of, from decay of metastable cadmium-115, quadrupole interactions of)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

effects in materials are given.

115<sub>In</sub>

IT 14336-64-2, properties RL: PRP (Properties)

(nuclear energy levels of, from silver-111 decay, quadrupole interactions of)

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

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L128 ANSWER 29 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1975:585169 HCAPLUS
ΑN
     83:185169
DN
OREF 83:29019a,29022a
ΤI
     Short-lived isotopes for use in nondestructive activation analysis
     Bode, P.; De Bruin, M.; Korthoven, P. J. M.
ΑU
CS
     Interuniv. React. Inst., Delft, Neth.
SO
     Journal of Radioanalytical Chemistry (1975), 26(1), 209-13
     CODEN: JRACBN; ISSN: 0022-4081
DT
     Journal
     English
LA
     71-8 (Nuclear Technology)
CC
     Section cross-reference(s): 79
     To obtain reliable data on short-lived isotopes for use in thermal neutron activation
AB
     anal., expts. were carried out by using a fast rabbit transfer system. Half-lives of 28
     short-lived isotopes were measured by using a counting system with a fixed dead-time. A
     Ge(Li) spectrometry system was used to determine the most important γ-ray energies and
     intensities of these isotopes. For the half-lives, an accuracy of better than 1% was
     attained, whereas for the \gamma-ray energies an accuracy of 0.1 keV was attained.
ΙT
     Radioelements, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (decay of, half-lives for)
ΙT
     Gamma ray
        (from radioelements of interest in activation anal.)
     Radiochemical analysis
TΤ
        (neutron activation, short-lived radioelement
        halflife for)
     14265-76-0, reactions 14265-77-1, reactions
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (decay of metastable, half-life for)
RN
     14265-76-0 HCAPLUS
     Hafnium, isotope of mass 179 (CA INDEX NAME)
CN
179<sub>Hf</sub>
RN
     14265-77-1 HCAPLUS
CN
     Hafnium, isotope of mass 178 (CA INDEX NAME)
 178<sub>Hf</sub>
     14191-71-0, reactions
                             14265-77-1, reactions
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (neutron capture by, γ-rays from)
RN
     14191-71-0 HCAPLUS
CN
     Indium, isotope of mass 115 (CA INDEX NAME)
115<sub>In</sub>
     14265-77-1 HCAPLUS
RN
     Hafnium, isotope of mass 178 (CA INDEX NAME)
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CN

178Hf

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L128 ANSWER 30 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1974:54701 HCAPLUS
ΑN
     80:54701
DN
OREF 80:8863a,8866a
     Gamma ray-multiplicity measurements for (\alpha,p), (\alpha,\alpha'),
ΤI
     (d,\alpha), (p,\alpha), (d,p), and (p,p') reactions
ΑU
     Degnan, J. H.; Cohen, B. L.; Rao, G. R.; Chan, K. C.; Shabason, L.
     Nucl. Phys. Lab., Univ. Pittsburgh, Pittsburgh, PA, USA
CS
SO
     Physical Review C: Nuclear Physics (1973), 8(6), 2255-66
     CODEN: PRVCAN; ISSN: 0556-2813
DT
     Journal
     English
LA
CC
     75-2 (Nuclear Phenomena)
AΒ
     By using a method introduced previously, measurements were made of the \gamma-ray multiplicity
     Ny (the average number of \gamma-rays emitted in the decay of residual nuclei left by nuclear
      reactions) as a function of excitation energy E* for (p,\alpha) and (d,\alpha) reactions on 51V,
      56Fe, and 57Fe targets; (\alpha, p) reactions on 51V, 56Fe, 57Fe, 58Ni, 64Ni, 93Nb, and Ag;
      (\alpha, \alpha') reactions on 51V, 55Mn, 56Fe, 57Fe, 59Co, and 64Ni; (d,p) reactions on 51V, 55Mn,
      56Fe, Ag, and 119Sn; and (p,p') reactions on 56Fe, 112Sn, and 122Sn. Bombarding energies
      ranged from 12 to 19 MeV. Ny was observed to be generally between 1 and 5 for E* between
      3 and 10 MeV for the (p,p'), (d,p), (d,\alpha), (p,\alpha), and (\alpha,\alpha') reactions investigated, and
      somewhat higher for the (\alpha,p) reactions. Ny increases with E*, and is larger for higher
      angular momentum transfer reactions although it is not as much larger as angular momentum
      transfer considerations alone would suggest. (This can be explained as a level-d. effect
      for compound-nucleus reactions- the average angular momentum of the states excited by the
      reactions is severely limited by the level-d. spin-cutoff parameter). The dependence of
     Ny on the average excess angular momentum of the residual nuclei was investigated by
     using calculated spin distributions for the residual nuclei. This dependence is similar
     to that for n capture Ny's-Ny increases with excess angular momenta for excess angular
     momenta of >2 or 3 units.
ΙT
     Gamma ray
        (from nuclear reactions, multiplicity of)
     14314-35-3, reactions
TΤ
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (deuteron bombardment of, multiplicity of γ-rays from)
RN
     14314-35-3 HCAPLUS
CN
     Tin, isotope of mass 119
119<sub>Sn</sub>
     10198-40-0, properties
TΤ
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (gamma rays from, from \alpha-particle bombardment of iron-57, multiplicity of)
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10198-40-0 HCAPLUS

Cobalt, isotope of mass 60

RN

CN

60co

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L128 ANSWER 32 OF 42 HCAPLUS COPYRIGHT ACS on STN
    1971:428388 HCAPLUS
ΑN
     75:28388
DN
OREF 75:4471a,4474a
TΙ
     Interpolative formulas for average nuclear level spacing and total
     radiation width
ΑU
     Musgrove, Anthony R.
CS
     Res. Establ., Aust. At. Energ. comm., Lucas Heights, Australia
     Aust. At. Energy Comm., AAEC/E [Rep.] (1970), AAEC/E211, 6 pp.
SO
     CODEN: AATCAZ
DT
     Report
LA
     English
CC
     75 (Nuclear Phenomena)
AΒ
     The free gas model formula for nuclear level ds. was used to interpolate unknown level
     spacings. The level d. parameter was fitted semi-empirically to give good fits to all
     exptl. measured level spacings. A formula for total radiation width is given which
     relates this quantity to the mass number, the average level spacing, and the excitation
     energy. Compound nuclei which have a long-lived isomeric state were found to have, on
     average, a radiation width 25 less than for non-isomeric compound nuclei. No correlation
     between radiation widths and n strength function was observed within the limits of
     accuracy of the exptl. data, contrary to some previous suggestions. Tables of calculated
     and measured resonance parameters are given for target nuclei in the range 60 < A < 203.
     The calculated 30 keV capture cross section is compared with measured and semiempirical
     data (Allen et al., 1970) and with the calculated values of Benzi and Reffo (1969).
ΙT
     Nuclear energy levels
        (spacing and width of, calcn. of)
ΙT
     Gamma rays
        (transitions of, calcn. of)
     14191-71-0, reactions
ΤТ
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (neutron capture cross sections and radiation width of, calcn. of)
     14191-71-0 HCAPLUS
RN
     Indium, isotope of mass 115 (CA INDEX NAME)
CN
115<sub>In</sub>
     10198-40-0, properties
ΙT
     RL: PRP (Properties)
        (nuclear energy level spacing and radiation width of, calcn. of)
RN
     10198-40-0 HCAPLUS
CN
     Cobalt, isotope of mass 60 (CA INDEX NAME)
```

60Co

```
L128 ANSWER 34 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1970:95576 HCAPLUS
ΑN
     72:95576
DN
OREF 72:17365a,17368a
ΤI
     Photon and electron excitation of nuclear isomers
ΑIJ
     Booth, Edward C.
CS
     Boston Univ., Boston, MA, USA
     U.S. Clearinghouse Fed. Sci. Tech. Inform., AD (1969),
SO
     AD-697153, 7 pp. Avail.: CFSTI
     From: U. S. Govt. Res. Develop. Rep. 1970, 70(2), 167
     CODEN: XCCIAV
DT
     Report
LA
     English
CC
     75 (Nuclear Phenomena)
AΒ
     Isomeric states of stable nuclei have been excited by photons and e via short-lived
     higher states. The isomer excitation expts. were undertaken in order to determine the
     relation between the total inelastic e scattering cross section and the radiative width,
     and to use the isomer excitation function to determine the energies widths, and angular
     momenta of accessible states. The following isotopes were studied: 87Sr, 103Rh, 111Cd,
     113In, 115In, 117Sn, 135Ba, 137Ba, 195Pt, and 199Hq.
     Gamma rays
ΙT
        (bombardment by, nuclear isomeric excitation in)
ΙT
     Nuclear energy levels
        (excitation of, by electrons and gamma rays)
     13981-59-4, properties 14191-71-0, properties
ΙT
     14191-88-9, properties
                              14336-64-2, properties
     RL: PRP (Properties)
        (nuclear energy levels of, excited by electrons and
        gamma rays)
RN
     13981-59-4 HCAPLUS
CN
     Tin, isotope of mass 117 (CA INDEX NAME)
117<sub>Sn</sub>
     14191-71-0 HCAPLUS
RN
CN
     Indium, isotope of mass 115 (CA INDEX NAME)
115<sub>In</sub>
RN
     14191-88-9 HCAPLUS
     Platinum, isotope of mass 195 (CA INDEX NAME)
CN
195<sub>Pt</sub>
RN
     14336-64-2 HCAPLUS
     Cadmium, isotope of mass 111 (CA INDEX NAME)
```

### L128 ANSWER 35 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1970:17446 HCAPLUS

DN 72:17446

OREF 72:3187a,3190a

- TI Nuclear photoactivation of selenium-77, silver-107, silver-109, cadmium-111, indium-115, and mercury-199
- AU Boivin, Michel; Cauchois, Yvette; Heno, Yvonne
- CS Lab. Chim. Phys., Fac. Sci. Paris, Paris, Fr.
- SO Nuclear Physics A (1969), 137(3), 520-30 CODEN: NUPABL; ISSN: 0375-9474
- DT Journal
- LA English
- CC 75 (Nuclear Phenomena)
- The photoactivation of various isotopes is studied by using the **x** -ray bremsstrahlung produced by a 2-MeV e accelerator. The photoactivation cross section was calculated for each of the observed nuclear levels. Some of these levels were observed for the 1st time. Their probable characteristics are proposed or confirmed. In certain cases it was possible to determine partial widths and the mean life of the level.
- IT X-rays

(bombardment by, of selenium-77 and silver isotopes)

IT 14191-71-0, properties 14336-64-2, properties

RL: PRP (Properties)

(nuclear energy levels of, from x-ray bombardment)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115<sub>In</sub>

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

### L128 ANSWER 36 OF 42 HCAPLUS COPYRIGHT ACS on STN

AN 1969:466270 HCAPLUS

DN 71:66270

OREF 71:12237a,12240a

### TI Photoactivation with a cobalt-60 source

AU Law, J.; Iddings, F. A.

CS Longwood Coll., Farmville, VA, USA

SO Journal of Radioanalytical Chemistry (1969), 3(1-2), 53-63 CODEN: JRACBN; ISSN: 0022-4081

DT Journal

LA English

CC 75 (Nuclear Phenomena)

AB Photoactivation of 13 elements was studied by using 5000-Ci. and 30,000-Ci. sources. Reported data from the study are presented in tables including exptl. conditions, observed activity, sensitivity achieved, . gamma. energies, and half-life observed, and reaction cross sections. Comparisons are made between this research and earlier reports.

IT 14191-71-0, analysis 14336-64-2, analysis RL: ANT (Analyte); ANST (Analytical study) (detection of, by gamma-ray activation)

RN 14191-71-0 HCAPLUS

CN Indium, isotope of mass 115 (CA INDEX NAME)

115<sub>In</sub>

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

#### **10/599,555** search, **12/4/09**, Page 54 of 72 L128 ANSWER 37 OF 42 HCAPLUS COPYRIGHT ACS on STN 1967:459835 HCAPLUS ΑN DN67:59835 OREF 67:11238e TΙ Excitation of metastable levels in nuclei by photoactivation on indium-gallium radiation loop ΑU Abrams, I.; Pelekis, L.; Taure, I. Neitronoakt. Anal. (1966), 135-40 SO CODEN: 16GJAI DTConference LA Russian 75 (Nuclear Phenomena) CC The isomers were obtained in an emitter of a radiation loop sphere-shaped and 15 cm. in AB diameter with a channel passing through it, 5 cm. in diameter An eutectic alloy (Sn 13, In 15, Ga 62%) was used as a circulating substance. The flux of $\gamma$ -quanta in the middle of the sphere was 4 + 1012 $\gamma$ -quanta/cm.2-sec. The following data were obtained: Isomer 111Cdm: $T1/2 = 47 \pm 1 \text{ min.}$ , $E\gamma = 152 \pm 4 \text{ and } 246 \pm 4 \text{ kev.}$ , cross section (6.9 ± 1.4) + 10-6 mb.; 115 Inm: $T1/2 = 4.23 \pm 0.08 \text{ hr.}$ , Ey = 335 ± 4 kev., cross section (1.5 ± 0.3) + Gamma rays ΙT (facilities, of gallium-indium alloy loop, excitation of metastable levels in) Nuclear energy levels ΤТ (metastable, photoactivation in gallium-indium alloy radiation loop) ΙT **14191-71-0**, properties **14336-64-2**, properties RL: PRP (Properties) (gamma-rays from and half-life of metastable) 14191-71-0 HCAPLUS RN

115<sub>Tn</sub>

CN

RN 14336-64-2 HCAPLUS CN Cadmium, isotope of mass 111 (CA INDEX NAME)

Indium, isotope of mass 115 (CA INDEX NAME)

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L128 ANSWER 38 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1964:408987 HCAPLUS
ΑN
     61:8987
DN
OREF 61:1436e-a
     Production of nuclear isomers. I. Production of nuclear isomers with
TΙ
     bremsstrahlung
ΑU
     Kaminishi, T.; Kojima, C.
     Nagoya Kogyo Gijutsu Shikensho Hokoku (1962), 11(5), 284-95
SO
     From: Phys. Abstr. 65(778), Abstr. No. 18271(1962).
     CODEN: NKGSAR; ISSN: 0027-7614
DT
     Journal
     Unavailable
LA
CC
     12 (Nuclear Phenomena)
     A number of stable elements were irradiated with bremsstrahlung from a 6-m.e.v. linear e
AB
     accelerator to produce directly each nuclear isomer which has been confirmed by other
     nuclear reactions. After irradiation the sample was moved to the counting system to
     measure the radioactivity produced and the isomeric nuclide was determined from the half-
     life and \gamma-ray energy of its decay. The production of the following isomers was
     confirmed: 77Sem, 79Brm, 87Srm, 89Ym, 103Rhm, 107Agm + 109Agm, 111Cdm, 113Inm + 115Inm,
     117Snm, 135Bam + 137Bam, 167Erm, 179Hfm, 183Wm, 191Irm, 195Ptm, 197Aum, and 199Hgm.
     Radioactive substances with halflives of 150 sec. in the V sample and 2.0 sec. in the W
     sample were produced, but not every radioactive nuclide could be determined because no
     information was available. Not all isomers of even-even nuclei were observed, although
     some half-lives and . gamma.-ray energies could be easily measured by this method if
     these isomers were produced.
ΙT
     Gamma rays
        (nuclear isomer production by)
     14191-88-9P, Platinum, isotope of mass 195 14265-76-0P, Hafnium, isotope of mass 179
ΙT
     RL: PREP (Preparation)
        (formation of metastable, by bremsstrahlung irradiation)
RN
     14191-88-9 HCAPLUS
     Platinum, isotope of mass 195 (CA INDEX NAME)
CN
195<sub>Pt</sub>
     14265-76-0 HCAPLUS
RN
CN
     Hafnium, isotope of mass 179 (CA INDEX NAME)
179<sub>Hf</sub>
TΤ
     14191-71-0P, Indium, isotope of mass 115
     RL: PREP (Preparation)
        (formation of metastable, by bremsstrahlung radiation)
     14191-71-0 HCAPLUS
RN
     Indium, isotope of mass 115 (CA INDEX NAME)
CN
115<sub>In</sub>
     13981-59-4P, Tin, isotope of mass 117
IT
     RL: PREP (Preparation)
        (production of metastable, by bremsstrahlung)
```

13981-59-4 HCAPLUS

Tin, isotope of mass 117 (CA INDEX NAME)

RN CN

117<sub>Sn</sub>

#### L128 ANSWER 39 OF 42 HCAPLUS COPYRIGHT ACS on STN 1953:54226 HCAPLUS AN 47:54226 DNOREF 47:9170f-q ΤI Angular correlation of nickel60 Wiedling, T. ΑU CS Univ. Stockholm Arkiv foer Fysik (1953), 7, 69-71 SO CODEN: AFYSA7; ISSN: 0365-2440 DTJournal LA English CC 3A (Nuclear Phenomena) AΒ A Co60 source in the form of CoCl2 gave an anisotropy, $(W(\pi)/W(\pi/2)] - 1$ , of 0.147 ± 0.007. Atomic nuclei ΙΤ (deuteron-bombarded Zn) ΙT Gamma rays (from cadmium-111, angular correlations of) ΙΤ **10198-40-0**, Cobalt, isotope of mass 60 (angular correlation of radiation from) 10198-40-0 HCAPLUS RN CN Cobalt, isotope of mass 60 (CA INDEX NAME) 60<sub>Co</sub>

14336-64-2, Cadmium, isotope of mass 111 ΙT (angular correlation of  $\gamma-\gamma$ cascade from)

RN 14336-64-2 HCAPLUS

CN Cadmium, isotope of mass 111 (CA INDEX NAME)

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L128 ANSWER 41 OF 42 HCAPLUS COPYRIGHT ACS on STN
     1950:55568 HCAPLUS
ΑN
DN
     44:55568
OREF 44:10539a-c
ΤI
     Radioactivity from enriched isotopes of cadmium
     Cork, J. M.; Rutledge, W. C.; Stoddard, A. E.; Branyan, C. E.; LeBlanc, J.
ΑU
CS
     Univ. of Michigan, Ann Arbor
     Physical Review (1950), 79, 938-9
SO
     CODEN: PHRVAO; ISSN: 0031-899X
DT
     Journal
     Unavailable
LA
CC
     3A (Nuclear Phenomena)
AΒ
     By irradiating in the pile specimens of Cd enriched in Cd108 and in Cd114, it is possible
     to interpret more positively the observed radioactivities and to evaluate the \gamma-rays
     associated with each. Cd109 produced by neutron capture in Cd108 decays with a half life
     of .apprx.250 days by K-capture to Ag109, in which a strong .gamma .-ray of 87.5 e.kv. is
     emitted. Cd115 is isomeric with half lives of 54 hrs. and 42.6 days, both isomers
     decaying to In115 by \beta-emission. The 54-hr. activity yields \gamma-rays of energies 335.5,
     343.7, 348.9, 369.3, 423.7, 451.9, 525.4, 559.1, and 713.1 e.kv. These \gamma-energies fit
     well a proposed nuclear level scheme in In115.
ΙT
     Gamma rays
        (from cadmium isotopes)
     Atomic nuclei
        (isomerism of Sn)
IΤ
     14191-71-0, Indium, isotope of mass 115
        (from decay of Cd115)
RN
     14191-71-0 HCAPLUS
CN
     Indium, isotope of mass 115 (CA INDEX NAME)
115<sub>In</sub>
     13981-59-4, Tin, isotope of mass 117
                                             14314-35-3,
ΤТ
     Tin, isotope of mass 119
        (isomeric)
     13981-59-4 HCAPLUS
RN
CN
     Tin, isotope of mass 117 (CA INDEX NAME)
117sn
     14314-35-3 HCAPLUS
RN
CN
     Tin, isotope of mass 119 (CA INDEX NAME)
119sn
```

10/599,555 search, 12/4/09, Page 58 of 72 L128 ANSWER 42 OF 42 HCAPLUS COPYRIGHT ACS on STN 1950:9465 HCAPLUS ΑN DN44:9465 OREF 44:1814b-e ΤI Coincidence arrangement for scintillation counter Meyer, K. P.; Baldinger, E.; Hahn, B.; Huber, P. ΑU CS Univ. Basel, Switz. SO Helvetica Physica Acta (1949), 22, 420-4 CODEN: HPACAK; ISSN: 0018-0238 DTJournal LA Unavailable CC 3A (Nuclear Phenomena) The apparatus consists of an anthracene crystal, two photomultiplier tubes, and a AB coincidence-mixing circuit containing Ge diodes. The background is reduced by a factor of 500 compared with that for a single photomultiplier counting at the rate of 10,000 per sec. The discrimination curve of the circuit gives a plateau so that the number of counts registered depends only on the source, crystal, and their relative positions and is independent of the photomultipliers and the amplification. In order to count coincident events, a second crystal and a third photomultplier tube are added to the circuit. With this arrangement and a Co60 source of 0.01 millicurie, measurements were made on the directional correlation of the two ;. gamma.-rays emitted in cascade. If K' and K'' are the number of true coincidences per min., at  $180^{\circ}$  and  $90^{\circ}$ , resp.,  $\eta = (K' K'')/K'' = 15 \pm 2\%$ . The background was 1 count per min., the space angle was 0.01, and the number of accidental and true coincidences per min. were 6 and 60, resp. ΙT Gamma rays (from cobalt-60) ΤТ Gamma rays (from indium-115) ΙT **14191-71-0**, Indium, isotope of mass 115  $(\beta$ - and  $\gamma$ -rays from) RN 14191-71-0 HCAPLUS CN Indium, isotope of mass 115 (CA INDEX NAME) 115<sub>In</sub>

ΙT

RN CN

60<sub>Co</sub>

10198-40-0, Cobalt, isotope of mass 60

Cobalt, isotope of mass 60 (CA INDEX NAME)

 $(\gamma$ -rays from) 10198-40-0 HCAPLUS

```
CAS/STN FILE 'REGISTRY' ENTERED AT 13:29:44 ON 04 DEC 2009
L1
           137 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?93NB?/CNS
            36 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?111CD?/CNS
L2
            27 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?113CD?/CNS
1.3
             6 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?135CS?/CNS
1.5
            62 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?115IN?/CNS
L6
           97 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?117SN?/CNS
1.7
            88 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?119SN?/CNS
L8
          106 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?125TE?/CNS
L9
            98 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?129XE?/CNS
L10
            61 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?131XE?/CNS
L11
            12 SEA FILE=REGISTRY SPE=ON
                                       ABB=ON PLU=ON
            9 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?179HF?/CNS
L12
            14 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?193IR?/CNS
L13
            89 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?195PT?/CNS
L14
    FILE 'STNGUIDE' ENTERED AT 13:30:02 ON 04 DEC 2009
    FILE 'REGISTRY' ENTERED AT 13:32:12 ON 04 DEC 2009
L15
           835 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON (L1 OR L2 OR L3 OR
               L4 OR L5 OR L6 OR L7 OR L8 OR L9 OR L10 OR L11 OR L12 OR L13 OR L14)
           773 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON (L1 OR L2 OR L3 OR
L16
               L4) OR (L6 OR L7 OR L8 OR L9 OR L10 OR L11 OR L12 OR L13 OR L14)
            62 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON ?115IN?/CNS
L17
    FILE 'STNGUIDE' ENTERED AT 13:32:13 ON 04 DEC 2009
    FILE 'HCAPLUS' ENTERED AT 13:34:27 ON 04 DEC 2009
           625 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L4
L18
            95 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L1
L19
          1298 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L2
L20
L21
          982 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L3
          1817 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5
L22
L23
           930 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L6
L24
          2387 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L7
L25
          1198 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L8
          2129 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L9
L26
L27
          1193 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L10
L28
           752 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L11
L29
           467 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON
L30
           524 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON
          1296 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L14
L31
    FILE 'REGISTRY' ENTERED AT 13:34:56 ON 04 DEC 2009
L32
           62 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 60CO?/CNS
    FILE 'HCAPLUS' ENTERED AT 13:34:57 ON 04 DEC 2009
L33
         10405 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L32
    FILE 'LCA' ENTERED AT 13:35:41 ON 04 DEC 2009
    FILE 'HCAPLUS' ENTERED AT 13:36:18 ON 04 DEC 2009
L34
             1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON US20080078961/PN
L35
               SEL PLU=ON L34 1- RN :
                                       14 TERMS
L36
           193 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L35
L37
             1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L34 AND L36
    FILE 'STNGUIDE' ENTERED AT 13:36:36 ON 04 DEC 2009
    FILE 'LCA' ENTERED AT 13:37:20 ON 04 DEC 2009
L38
          1992 SEA FILE=LCA SPE=ON ABB=ON PLU=ON GAMMA
L39
            96 SEA FILE=LCA SPE=ON ABB=ON PLU=ON CASCAD?
L40
             O SEA FILE=LCA SPE=ON ABB=ON PLU=ON RADIOCASCAD?
L41
           677 SEA FILE=LCA SPE=ON ABB=ON PLU=ON DECAY?
            6 SEA FILE=LCA SPE=ON ABB=ON PLU=ON HALFLI?
L42
L43
           230 SEA FILE=LCA SPE=ON ABB=ON PLU=ON HALF LIFE
            57 SEA FILE=LCA SPE=ON ABB=ON PLU=ON HALF LIVES
T.44
L45
           200 SEA FILE=LCA SPE=ON ABB=ON PLU=ON (LIFE OR LIVE##)(2A)(SHORT####### OR LONG####### OR
LENGTH######)
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CAS/STN (CONTINUED)
    FILE 'LCA' (CONTINUED)
             O SEA FILE=LCA SPE=ON ABB=ON PLU=ON LONGLIF?
             1 SEA FILE=LCA SPE=ON ABB=ON PLU=ON LONGLIV?
L48
             O SEA FILE=LCA SPE=ON ABB=ON PLU=ON SHORTLIV?
1.49
             O SEA FILE=LCA SPE=ON ABB=ON PLU=ON SHORTLIF?
            27 SEA FILE=LCA SPE=ON ABB=ON PLU=ON DEEXCIT? OR DE EXCIT?
L50
          1573 SEA FILE=LCA SPE=ON ABB=ON PLU=ON EXCIT?
T.51
L52
           160 SEA FILE=LCA SPE=ON ABB=ON PLU=ON (MIX##### OR BLEND##### OR PAIR#### OR DOUBLE OR
TWO OR SECOND OR 2ND OR COUPL#### OR ANOTHER) (3A) (?ISOMER? OR ?NUCLIDE? OR NUCLEIDE? OR ?SPECIES?)
            47 SEA FILE=LCA SPE=ON ABB=ON PLU=ON (MIX##### OR BLEND#####
                OR PAIR##### OR DOUBLE OR TWO OR SECOND OR 2ND OR COUPL#### OR ANOTHER) (3A) ?ISOTOP?
             24 SEA FILE=LCA SPE=ON ABB=ON PLU=ON ENTANGL######
L54
    FILE 'HCAPLUS' ENTERED AT 13:42:46 ON 04 DEC 2009
L55
           73 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L22 AND L33
           115 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L18 OR L19 OR L20 OR L21) AND L33
L56
           153 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L23 OR L24 OR L25 OR L26 OR L27 OR L28 OR L29
OR L30 OR L31) AND L33
            236 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L23 OR L24 OR L25 OR
               L26 OR L27 OR L28 OR L29 OR L30 OR L31) AND L22
            230 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L18 OR L19 OR L20 OR L21) AND L22
L59
    FILE 'STNGUIDE' ENTERED AT 13:42:48 ON 04 DEC 2009
    FILE 'HCAPLUS' ENTERED AT 13:45:05 ON 04 DEC 2009
L60
           142 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L58 OR L59) AND
                (GAMMA OR X(2A) (RAY OR RADIATION OR PHOTON######))
             73 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L22 AND L33
L61
L62
           170 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L56 OR L57) NOT L55
           368 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L60 OR L61 OR L62)
L63
L64
           197 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L38
             9 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L39
L65
L66
             0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L40
            65 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L41
L67
L68
            2 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L42
L69
           20 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L43
L70
           13 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L44
           31 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L45
0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L46
0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L47
L71
L72
L73
            0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L48
L74
            O SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L49
L75
            1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L50
L76
L77
           59 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L51
            6 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L52
L78
L79
            3 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L53
L80
            1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L54
L81
               SEL PLU=ON L34 1- IPC ECLA FTERM NCL : 5 TERMS
           6074 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L81
L82
L83
             1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L63 AND L82
    FILE 'STNGUIDE' ENTERED AT 13:48:04 ON 04 DEC 2009
     FILE 'HCAPLUS' ENTERED AT 13:50:07 ON 04 DEC 2009
             19 SEA FILE-HCAPLUS SPE-ON ABB-ON PLU-ON L63 AND X(2A) (RAY OR
L84
                RADIATION OR PHOTON######)
             45 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L64 AND L67
L85
1.86
             53 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L64 AND L77
             17 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L67 AND L77
L87
     FILE 'LCA' ENTERED AT 13:50:49 ON 04 DEC 2009
     FILE 'HCAPLUS' ENTERED AT 13:52:25 ON 04 DEC 2009
1.88
           135 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L65 OR (L68 OR L69 OR
                L70 OR L71 OR L72 OR L73 OR L74 OR L75 OR L76) OR (L78 OR L79
               OR L80) OR L83 OR (L84 OR L85 OR L86 OR L87)
L89
           135 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L88 NOT L34
```

53 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L89 AND 1940-1980/PY

L90

#### CAS/STN (CONTINUED) FILE 'HCAPLUS' (CONTINUED) 74 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L89 AND 1981-2005/PY L92 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L89 AND 1981-2004/PRY L93 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L89 AND 2005/PRY AND (WO OR US)/PRC 1.94 127 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L90 OR L91 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND ENTANGL########(6A)GAMMA L95 480 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON ENTANGL######## AND GAMMA L96 L97 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L96 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND ENTANGL? 3 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND (MIX##### OR BLEND##### OR PAIR##### OR L98 L99 DOUBLE OR TWO OR SECOND OR 2ND OR COUPL#### OR ANOTHER) (3A) ?ISOTOP? L100 6 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND (MIX##### OR BLEND##### OR PAIR##### OR DOUBLE OR TWO OR SECOND OR 2ND OR COUPL#### OR ANOTHER) (3A) (?ISOMER? OR ?NUCLIDE? OR NUCLEIDE? OR ?SPECIES?) I.101 41 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)GAMMA 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)CASCAD? L102 7 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5(L)CASCAD? L104 O SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)?CO60? 2 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)?60CO? 0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)CO L105 L106 O SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L5(L)COBALT 4 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5(L)COBALT 5 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L5(L)CO 73 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L22 AND L33 30 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L110 AND GAMMA L107 T-108

FILE 'LCA' ENTERED AT 13:58:08 ON 04 DEC 2009

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L109 L110 L111

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FILE 'HCAPLUS' ENTERED AT 13:59:44 ON 04 DEC 2009
            121 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L65 OR L68 OR L70 OR
L112
                 L76 OR (L78 OR L79 OR L80) OR L84 OR L87 OR (L99 OR L100 OR
                 L101 OR L102 OR L103 OR L104 OR L105 OR L106 OR L107 OR L108 OR L109) OR L111
L113
               1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L94 AND L36
L114
              1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L113 NOT L34
L115
            122 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L112 OR L113 OR L114) NOT L34
L116
            64 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L115 AND 1980-2005/PY
L117
             50 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L115 AND 1940-1979/PY
           114 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L116 OR L117)
1 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND DEEXCIT?
0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND DE EXCIT?
L118
L119
L120
             10 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND EXCIT#######(2A)STATE
L121
            20 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND LIFE
L122
            20 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND LIVE##
L123
            2 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND HALFLI?
0 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND SHORTLI?
L124
L125
             O SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND LONGLI?
L126
L127
              6 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L118 AND (GAMMA OR X)(5A)CASCAD?
            42 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON L114 OR (L119 OR L120
L128
                 OR L121 OR L122 OR L123 OR L124 OR L125 OR L126 OR L127)
```

### EAST Search History for a search on 12-4-2009:

Search History for a search on 12-4-2009:

13 Public 19 Public 19

Ref #	Hits	Search Query	DBs	Defa ult Oper ator	Plurals	Time Stamp
L1	0	"2008315210"	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 08:08
L2	1	*20080315210"	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 08:08
Ľ3	1	"20030078961"	EPO; IPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 08:29
L4	1	2005-716517.NRAN.	DERWENT	OR	OFF	2009/11/30 08:33
<b>ι</b> Ζ/	Ō	(mix\$6 or blend\$5) near5 two adj1 (isotopes or radioisotopes or isomers or radioisomers) and (half adj lif\$2 or half adj liv\$2)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	-2009/11/30 09:37
L8	0	(mix\$6 or blend\$5) near5 two adj1 (isotopes or radioisotopes or isomers or radioisomers) and decay\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:41
L9	21 	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and decay\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON:	2009/11/30 09:41
L10	0	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and halflif\$4	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:41
Ļ11	o.	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and haifliv\$4	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:41
L12	4	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and half adj liv\$4	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:42
13	22	(mix\$6 or blend\$5) near5 (isotopes or radioisotopes or isomers or radioisomers) and half adj lif\$4	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:42
L14	42	L9 L12 L13	EPO; JPO; DERWENT; IBM_TD8	OR	ON	2009/11/30 09:42
L.15	1	L14 and shift\$6	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:43
L16	1	L14 and curv\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:43

L17	2	L14 and g21k\$	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:44
1.18	3	L14 and R09-e\$	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:44
L19	5	L17 L18	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:44
L20	29639	g21k <b>\$</b>	EPO; JPO; DERWENT; JBM_TDB	OR	ON	2009/11/30 09:50
L21	4010	k09~e\$	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L22	33573	120121	EPO; JPO; DERWENT; JBM_TDB	CR	ON	2009/11/30 09:51
L23	3	L22 and decay\$5 near5 shorter	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
124	. 12	L22 and decay\$5 near5 short\$8	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L25	7	L22 and decay\$5 near5 long\$5	EPO; JPO; DERWENT; IBM_TDB	OR	OM	2009/11/30 09:51
L26	0	L22 and decay\$5 near5 length\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 09:51
L27	1	L22 and decay\$5 near5 curv\$6	EPO; JPO; DERWENT; IBM_TDB	OR	OM	2009/11/30 09:52
128	.19	L23 L24 L25 L27	EPO; JPO; DERWENT; IBM_TDB	OR	ON:	2009/11/30 09:52
L29	5	L28 and gamma	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:22
130	152	gamma near7 decay\$7	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:24
L31	0	£30 and (mix\$6 or blend\$6) near4 two	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:25

L32	1	L30 and (mix\$6 or blend\$6 or two or second) near2 (isomer\$5 or radioisomer\$5 or radioisotop\$6 or isotop\$6 or nuclid\$5 or nucleid\$6 or radio or radionucl\$9)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:26
1.33	0	L30 and composite near3 half	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:27
L34	18	L30 and lives	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:28
135	18	1.30 and lifes	EPO; JPO; DERWENT; JBM_TDB	OR	ON	2009/11/30 10:28
L36	14	L30 and half	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:28
L37:	19	L34.L35.L36	EPO; JPO; DERWENT; IBM_TDB	OR .	ON	2009/11/30 10:28
L38	0	L37 and composite	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L39	0	L37 and mlx\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON .	2009/11/30 10:29
L40	0	L37 and blend\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L41	6	L37 and two	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L42	4	L37 and second	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:29
L43	10		क्षीत्राच व्यक्तप्रकारिक	OR	ON	2009/11/30 10:29
L44	1178	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 half adj life	EPO; JPO; DERWENT	OR	ON	2009/11/30 10:32
L45	31	A CONTRACTOR OF THE PROPERTY O	EPO; JPO; DERWENT IBM_TDB	OR	ON	2009/11/30 10:32

L46	31	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 half adj lives near6 (composite or blend\$6 or mix\$6 or effective)	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:32
L47	0	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 halflives near6 (composite or blend\$6 or mix\$6 or effective)	EPO; JPO; DERWENT; JBM_TDB	OR	ON	2009/11/30 10:33
Ł48	0	(vary\$6 or vari\$5 or chang\$6 or increas\$5 or decreas\$6) near6 halflife near6 (composite or blend\$6 or mix\$6 or effective)	EPO; JPO; DERWENT; IBM_TD8	OR	ON Constitution of the Con	2009/11/30 10:33
L49	31	145146	EPO; IPO; DERWENT; IBM_TDB	OR	ON st	2009/11/30 10:33
L50	0	L49 and decay\$5 near4 curv\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L51	0	1.49 and decay\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L52	0	L49 and curv\$5	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:33
L53	0	L49 and gamma	EPO, JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L54	0	L49 and skew\$6 near6 effective adj half adj life	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L55	0	skew\$6 near6 effective adj half adj life	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L56	2	graph\$5 near6 effective adj half adj life	EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:34
L57	0	curv\$6 near6 effective adj half adj life	EPO; JPO; DERWENT; JBM_TDB	OR	ÖN 11	2009/11/30 10:34
L58	0	skew\$6 near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:35

L59	1	curv\$5 near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TD8	OR	ON	2009/11/30 10:35
L60	4	graph\$6 near6 effective ad) half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:36
L61	O	fig near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:36
L62	0	figure near6 effective adj half adj life	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TOB	OR	ON	2009/11/30 10:36
L63	5	L56 L59 L60	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:36
L64	6862	(long\$6 near7 short\$5) near7 (life or lived or live or halfilfe or halfilve\$1)	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:44
L65	10	(long\$6 near7 short\$5) near7 (life or lived or live or halflife or halflive\$1) near7 (isptop\$5 or radioisotop\$6 or nuclide or radionu\$8 or nucleide or isomer\$6 or radioisomer\$6) near7 (two or second)	US-PGPUB; USPAT; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:46
L67	O	hybrid near4 radiation near4 source : near12 (decay\$6 or long\$5 or short\$6 or half or halfii\$6) near12 gamma	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; JBM_TDB	OR	ON	2009/11/30 10:54
L68	0	hybrid near4 (radiation or source) near12 (decay\$6 or long\$5 or short\$6 or half or halfli\$6) near12 gamma	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:55

L69	47	hybrid near4 radiation near4 source	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR:	ON	2009/11/30 10:55
L70	.0	L69 and g21k\$6	US-PGPUB; USPAT; USOGR; EPO; JPO; DERWENT; IBM_TDB	OK	ON	2009/11/30 10:57
L71	2	(nix\$7 or blend\$6) near4 halflife	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:57
L72	36	(hix\$7 or blend\$6) near4 haif adj life	US PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 10:57
L73	36	£71 L72	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2909/11/30 10:58
174	10	L64 and g21k\$	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:00
L75	56	"5674177"	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:06
L76	25	L75 and (long\$5 or short\$6) near5 (life or lives or halfli\$6 or lived or decay\$6)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; JBM_TDB	OR	en	2009/11/30 11:07

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L77	10	L75 and (relat\$4 or relation\$7 or relativ\$5 or compar\$4 or comparat\$6) near5 (life or lives or halfii\$6 or lived or decay\$6)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:09
L78	10	L76 and L77	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	an	2009/11/30 11:09
L79	1008	(curv\$5 or graph\$5 or fig or figure) near4 (half ad) life)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:12
1.80	0	(cury\$5 or graph\$5 or fig or figure) near7 ((haif adj life) or halfli\$7) near7 (hybrid or composite or mix\$3 or mixture or blend\$5 or two) near7 (radio or radioactiv\$6 or radioisotop\$6 or nuclide or nucleide or radionu\$7 or radiolsomer\$6 or isomer\$5)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:15
L81	89	(curv\$5 or graph\$5 or fig or figure) same ((half adj life) or halfli\$7) same (hybrid or composite or mix\$3 or mixture or blend\$5 or two) near7 (radio or radioactiv\$6 or radioisotop\$6 or nuclide or nucleide or radionu\$7 or radioisomer\$6 or isomer\$5)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:16
L82	20	£81 and shorter	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:16
L83	32	L81 and longer	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB		ON	2009/11/30 11:17

L86	43	L81 and gamma neAR5 (EMIT\$6 OR EMISS\$5 or decay\$5)	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:18
£87	8.	L81 and L82 and L83 and L84	USPAT; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2009/11/30 11:19
L88	O	"3017514" and halflives	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2009/11/30 11:21
L89		"3017514" and halflifes	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2009/11/30 11:21
L90	O	"3017514" and half adj lifes	US-PGPU8; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2009/11/30 11:21
191	1	"3017514" and half adj lives.	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	OFF	2009/11/30 11:21
192	3	"3017514" and (composite or mix\$3 or mixture or blend\$6 or composition or compos\$4 or compris\$6) near3 two	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	***************************************	ON	2009/11/30 11:23



# **VOLUNTARY SEARCH FEEDBACK**



Art Unit	App./Serial #		
How did you use	your search results?	You may cut and paste into the box b	elow
☐ 102 rejection ☐ 103 rejection ☐ Cited in allowar	Citations or Patents Used	acion #, author, or patent #	
		t in technology, or specific invention	
Kesults verified	the lack of relevant prior	art (helped determine patentability).	
,	Types Patent(s)	Non-Patent Literature	
COMMENTS			
	Questions about the so	cope or the results of the search?	
	•	searcher or EIC Supervisor.	
	Please submit co	ompleted form to your EIC.	
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د سوم مراه می است.	MANAGEMENT OF THE STATE OF THE		
Additional Notes if appl	licable (please indicate all actions	s including emails, phone calls, and individuals assisting):	
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